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### Concise Asymmetric Total Synthesis of Scyphostatin, a Potent Inhibitor of Neutral Sphingomyelinase

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Abstract: The concise asymmetric total synthesis of scyphostatin has been achieved by condensation of the optically active cyclohexane unit, prepared from the commercially available 1,4-cyclohexadiene by our own method, and the side chain, prepared by the method developed by Hoye and Tennakoon (T. R. Hoye, M. A. Tennakoon, [Org.](http://dx.doi.org/10.1021/ol0058386) Lett. **2000**, 2, 1481-1483). The modification of the epoxy cyclohexenone unit was achieved in a late stage of the total

#### Introduction

Scyphostatin  $(1)^{[1]}$  was isolated from Dasyscyphus mollissimus SANK-13892 in 1997 and showed the most potent activity  $(IC_{50} = 1.0 \mu M)$  toward a neutral sphingomyelinase (N-SMase) among the large number of already reported N-SMase inhibitors.[2] N-SMase is the enzyme that accelerates the conversion of sphingomyelin into ceramide. Since the ceramide is believed to be an intracellular lipid second messenger and to play vital roles in the regulation of cell proliferation, modulation of the inflammatory process, and apoptosis, $^{[3]}$  1 is recognized as a lead compound for the treatment of inflammation and autoimmune diseases. This molecule consists of an epoxy cyclohexenone unit with an amino alcohol moiety and an unsaturated fatty acid side chain, so that 1 is a challenging target compound for many organic chemists in view of not only its biological activity but also its unique structure. Although several synthetic studies toward

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synthesis, and deprotection of the primary alcohol was conducted in the final step. During the synthesis several key reactions were attained: 1) intramolecular bromoetherification of the cyclohexadiene acetal; 2) stereoselec-

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tive introduction of the tertiary alcohol, 3) deprotection of the acetal function to the aldehyde by combination with silyl triflate/2,4,6-collidine and the one-pot synthesis of the disilyl aldehyde compounds, with different types of silyl groups, from the dihydroxy acetal compounds; and 4) facile deprotection of the 2,4-dimethoxyphenylmethyl  $(^{2,4}DMPM)$  protecting group of the primary alcohol.



1 have already been reported,  $[4, 5]$  to the best of our knowledge only two total asymmetric syntheses of 1 have, quite recently, been reported by Katoh and co-workers from Darabinose derivatives in  $22 \text{ steps}^{[6]}$  and by Takagi et al. through a spirolactone derivative from tyrosine in 18 steps.[7]

We now present our study on the total asymmetric synthesis of 1. For the optically active epoxy cyclohexenone unit of 1, we recently succeeded in the synthesis of model compound 5, which has the same epoxy cyclohexenone unit of 1.<sup>[8]</sup> Thus the intramolecular bromoetherification of the diene acetal 2 gave a cyclohexene eight-membered acetal 3 in a stereoselective manner.[9] Compound 3 was transformed into the allyl alcohol 4, which was further modified to 5. In the beginning, we attempted to convert the dimethyl acetal function of 5 into an aldehyde function, which would be used for the further construction of the amino alcohol moiety and the introduction of the fatty acid side chain. However, the conversion of 5 into 6 was unsuccessful, possi-

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bly as a result of the instability of the epoxy cyclohexenone unit (Scheme 1). We then decided to develop an alternative way to synthesize 1. As another synthetic plan, we had to consider the instability of 1, as it is a very unstable com-



Scheme 1. Attempted conversion of  $5$  into 6. NBS = N-bromosuccinimide.

pound under acidic and basic conditions. For example, the epoxy cyclohexenone unit of 1 has many reactive functional groups, such as the enone, unstable under basic conditions, and the epoxide, unstable under acidic conditions. The primary alcohol of 1 causes an intramolecular nucleophilic attack on the ketone under basic conditions. Furthermore, the trienamide unit of the side chain causes decomposition even under neutral conditions.<sup>[1b]</sup>

A summary of our total synthesis is depicted in Scheme 2. We first condensed the optically active cyclohexane unit, prepared by our own method, and the side chain prepared by the method developed by Hoye and Tennakoon.<sup>[5a]</sup> Then modification of the epoxy cyclohexenone unit was achieved in a late stage of the total synthesis, and deprotection of the primary alcohol was conducted in the final step. We then completed the first generation asymmetric total synthesis of 1. We next improved two reactions, thus giving a more efficient second generation asymmetric total synthesis of 1 in a total of 17 steps.



Scheme 2. Summary of the synthesis of scyphostatin (1).

#### Results and Discussion

Examination of the synthetic route to an epoxy cyclohexenone unit with an amino alcohol moiety: Herein, the studies of the synthetic route to an epoxy cyclohexenone unit with an amino alcohol moiety using cyclohexene eight-membered acetal 4 are described. In a previous study, $[9]$  the cyclohexadiene acetal 2 was prepared by the transacetalization of diethylacetal 8, obtained from benzoic acid by using a fourstep sequence according to a previously reported procedure<sup>[10]</sup> with  $(R, R)$ -hydrobenzoin. We then succeeded in synthesizing 8 in only one step. Thus, lithiation of commercially available 1,4-cyclohexadiene (7) and successive reaction with bromoacetaldehyde diethyl acetal afforded 8 in a onepot operation in 75% yield. The transformation of 8 into dimethyl acetal 10 was carried out by our previously reported procedure:<sup>[8]</sup> 1) Transacetalization of **8** with  $(R, R)$ -hydrobenzoin, 2) intramolecular bromoetherification of 2 with NBS in the presence of MeOH, 3) radical reduction of 3, 4) stereoselective oxidation of 9 by using  $SeO<sub>2</sub>$ , and 5) acidic treatment of 4 in MeOH (Scheme 3).



Scheme 3. Synthesis of cyclohexene dimethyl acetal 10. i) sec-BuLi, TMEDA, THF,  $-78 \text{°C}$ , then BrCH<sub>2</sub>CH(OEt)<sub>2</sub>; ii) (R,R)-hydrobenzoin,  $p$ -TsOH, toluene, 50°C; iii) Bu<sub>3</sub>SnH, AIBN, benzene, reflux; iv) SeO<sub>2</sub>, pyridine, dioxane, 70°C; v) cat. PPTS, MeOH, RT. AIBN = azobisisobutylonitrile, PPTS=pyridinium p-toluene sulfonate, TMEDA= $N, N, N', N'$ tetramethylethylenediamine, p-TsOH=p-toluenesulfonic acid.

We then examined the construction of the amino alcohol moiety. First, we prepared the p-methoxyphenylmethyl  $(MPM)$  ether 14a from 10 by using a three-step sequence (Scheme 4, a series): 1) Methylation of the benzyl alcohol, 2) acid hydrolysis of the acetal, and 3) nucleophilic addition of 4-methoxyphenylmethyloxymethyl lithium from the stannyl compound  $13$ .<sup>[11]</sup> The stereochemistry of the secondary alcohols 14a and 14a' was determined by using a modified Mosher's method (see the Experimental Section).<sup>[12]</sup> In this case, the undesired S alcohol 14 a' was obtained as a major product. Our study using 14a is shown in Scheme 5. The Mitsunobu reaction<sup>[13]</sup> of the secondary alcohol **14a** using DPPA for introduction of the azide group with inversion was unsuccessful and gave a diastereomeric mixture of the allylic azide 15 because of the presence of the reactive allylic tertiary alcohol. We then attempted to first convert the olefin and then introduce a nitrogen atom. In this case, the stereoselective epoxidation<sup>[14]</sup> of **14a** with  $[VO(acac)_2]$  and TBHP proceeded well without any problems to give the epoxy alcohol 16. However, the following Mitsunobu reaction afforded a low yield of the desired azide 17 accompa-



Scheme 4. Syntheses of MPM ethers 14a and 14b. i) MeI, NaH, THF/ DMF (1:1),  $0^{\circ}C$ ; ii) HCO<sub>2</sub>H (10%), CH<sub>3</sub>CN (90% aq.), RT for 12a or TESOTf, 2.6-lutidine, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C for  $12b$ ; iii) MPMOCH<sub>2</sub>SnBu<sub>3</sub> (13), n-BuLi, THF,  $-78$ °C. DMF=N,N-dimethylformamide, TESOTf=triethylsilyl trifluoromethanesulfonate.



Scheme 5. Introduction of a nitrogen atom into  $14a$ . i) DPPA, PPh<sub>3</sub>, DEAD, THF, RT; ii) TBHP,  $[VO(acac)_2]$ , toluene,  $0^{\circ}C$ ; iii) DPPA, PPh<sub>3</sub>, DEAD, THF, RT; iv) MsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0°C; v) NaN<sub>3</sub> or TMSN<sub>3</sub>. DEAD=diethyl azodicarboxylate, DPPA=diphenylphosphorylazide, Ms=methanesulfonyl, TBHP=tert-butylhydroperoxide, TMS=trimethylsilyl.

nied by the allylic azide 18, which was obtained by  $\beta$ -elimination of the tertiary alcohol 17. In another trial using 14a and 16, the selective mesylation of the secondary alcohol followed by nucleophilic substitution resulted in low yields of the mixture of the epimeric azide isomers C and D to our disappointment. Furthermore, the reduction of the azide group of 17 did not afford the desired amino compound possibly as a result of the presence of the tertiary alcohol.

Total Synthesis of Scyphostatin **FULL PAPER** 

Since the presence of the naked tertiary alcohol caused unfavorable side reactions, as shown in Scheme 5 (a series), we next examined a route using the triethylsilyl (TES) ether 14b, in which the tertiary alcohol was protected as a silyl ether. Compound 14b was prepared from 10 almost in the same procedure as 14a shown in Scheme 4. Thus, the deprotection of the acetal function and silylation of the tertiary alcohol of  $11$  by using our recently developed method,  $[15]$  a combination of TESOTf and 2,6-lutidine followed by a work-up with water gave the TES ether aldehyde 12b in excellent yield. The subsequent nucleophilic addition of the lithiated 13 gave the diastereomixture of 14b and 14b'. The stereochemistries of the secondary alcohols were also determined by the modified Mosher's method.<sup>[12]</sup> To our delight, the ratio of the two alcohols 14b and 14b' was reversed compared to the ratio of 14a and 14a', and the desired alcohol 14b was obtained as the major product. This outcome shows that the silylation of the tertiary alcohol increases the ratio of the desired alcohol.

Introduction of a nitrogen atom and its conversion are shown in Scheme 6. The Mitsunobu reaction of 14b gave the azide 19 in good yield. Formation of an undesirable allylic azide, such as 15, was not observed. For examination of the synthetic route available for the synthesis of 1, a benzoyl group was introduced in place of the unsaturated fatty acid side chain. The reduction of 19 using  $LiAlH<sub>4</sub>$  followed by Nacylation gave the benzoate 20. Deprotection of the MPM and TES ethers gave the diol 21 in two steps. Protection of the primary alcohol with a tert-butyldimethylsilyl (TBS), tert-butyldiphenylsilyl (TBDPS), or acetyl (Ac) group, respectively, and stereoselective epoxidation of each compound gave the epoxy alcohols 22 a–c. However, the reactions to remove the hydrobenzoin unit of 22a–c caused several side reactions, such as opening of the oxirane ring, cleavage of the amide bond, deprotection of the primary alcohol, and so on, thus the yields of the desired compounds 23 a–c were low. These results showed that the removal of the hydrobenzoin unit is necessary at an early stage.



Scheme 6. Introduction of a nitrogen atom into  $14b$ . i) DPPA, PPh<sub>3</sub>, DEAD, THF, RT; ii) LiAlH<sub>4</sub>, THF, 0°C; iii) BzCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, RT; iv) DDQ, CH<sub>2</sub>Cl<sub>2</sub>, RT; v) TBAF, THF, RT; vi) TBSCl, imidazole, DMF, RT for 22a; TBDPSCl, imidazole, DMF, RT for 22b; Ac<sub>2</sub>O, pyridine, RT for 22c; vii) TBHP,  $[VO(acac)_2]$ , toluene, RT; viii) 1.  $[Pd(OH<sub>2</sub>)]$ , H<sub>2</sub>, MeOH, RT, 3.5 atm or 2. liq. NH<sub>3</sub>, ca.  $-78$  °C. acac=acetylacetonate, DDQ=2,3-dichloro-5,6-dicyano-1,4-benzoquinone, TBAF=tetra-n-butylammonium fluoride, TBDPSCl=tert-butyldiphenylsilyl chloride, TBSCl=tert-butyldimethylsilyl chloride.

Asymmetric total synthesis of 1; first generation: Based on the results described above, we decided to remove the hydrobenzoin unit before constructing the oxirane ring, forming the amide bond, and deprotecting the primary alcohol. Since the epoxy cyclohexenone unit is considered to be unstable, in other words, very reactive under various reaction conditions, its construction was carried out at a late stage of the total synthesis. Our first generation asymmetric synthesis of 1 is shown in Scheme 7. Birch reduction of the hydrobenzoin unit of 10, prepared from 1,4-cyclohexadiene (7) in six steps (Scheme 2), gave the diol 24, which was treated with TBSCl/imidazole to afford the hydroxy silyl ether 25. Deprotection of the acetal function of 25 by our recently developed method<sup>[15]</sup> gave the disilylated aldehyde 26 a in high yield. Nucleophilic addition of 4-methoxyphenylmethyloxymethyl lithium from the stannyl compound 13 to 26 a gave the two alcohols  $(R)$ -27 a and  $(S)$ -27 a' in a ratio of 2.2:1 and 81% total yield. The stereochemistry of the two alcohols was determined by the modified Mosher's method<sup>[12]</sup> (see the Experimental Section). The desired R isomer 27 a was converted into amine 29 a through azide **28a** by the Mitsunobu reaction<sup>[13]</sup> in 69% yield followed by reduction using LiAlH4. Condensation of the amine 29 a and



Scheme 7. First generation asymmetric total synthesis of scyphostatin (1). i) Ca, EtOH, liq. NH<sub>3</sub>,  $-40^{\circ}$ C; ii) TBSCl, imidazole, DMF, RT; iii) TESOTf, 2,4,6-collidine, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, then H<sub>2</sub>O; iv) Bu<sub>3</sub>SnCH<sub>2</sub>OMPM  $(13)$ , nBuLi, THF,  $-78^{\circ}$ C; v) DPPA, PPh<sub>3</sub>, DEAD, CH<sub>2</sub>Cl<sub>2</sub>, RT; vi) LiAlH<sub>4</sub>, THF, RT; vii) 30, DCC, DMAP,  $CH_2Cl_2$ ,  $0^{\circ}C$ ; viii) TBAF, THF,  $0^{\circ}C$ ; ix) TBHP,  $[VO(acc)_2]$ , toluene,  $0^{\circ}C$ ; x) TPAP, NMO, CH<sub>2</sub>Cl<sub>2</sub>,  $0^{\circ}C$ ; j) LDA, [15]crown-5, Ph(Cl)S=NtBu, THF,  $-78\text{°C}$ ; xii) Ph<sub>3</sub>C<sup>+</sup>BF<sub>4</sub><sup>-</sup>, CH<sub>2</sub>Cl<sub>2</sub>, 0°C. DCC=N,N'-dicyclohexylcarbodiimide,  $DMAP =$ dimethylaminopyridine,  $LDA =$ lithium diisopropylamide,  $NMO = N$ -methylmorpholine- $N$ -oxide, TPAP = tetra-n-propylammonium perruthenate.

side chain acid 30, prepared by the method developed by Hoye and Tennakoon,<sup>[5a]</sup> gave amide  $31a$ , which was desilylated to give diol 32 a in 47% yield from azide 28 a. Stereoselective epoxidation<sup>[14]</sup> of **32a** with  $[VO(acac)_2]$  and TBHP gave *cis*-epoxy alcohol 33 a in 69% yield as a single isomer. Oxidation of the secondary alcohol of  $33a$  using TPAP/  $NMO^{[16]}$  gave the ketone 34a in 49% yield (88% based on the consumed  $33a$ ). The treatment of  $34a$  with LDA and the reaction with N-tert-butylphenylsulfinimidoyl chloride gave enone 35 a in 37% yield (51% based on consumed 34a).<sup>[17]</sup> The addition of  $[15]$ crown-5 to the reaction mixture improved the yield. Deprotection of the MPM ether to give the alcohol was successfully done using trityl tetrafluoroborate  $(\text{Ph}_3\text{C}^+\text{BF}_4^-)$  to afford scyphostatin (1) in 32% yield,<sup>[18]</sup> whereas CAN and DDQ, usually used for the deprotection of the MPM ether.<sup>[19]</sup> gave a complex mixture as a result of the reaction at the trienamide unit (the same results were obtained in our model study of 35 a with a trienamide unit, as described later in Table 2). The spectroscopic data of synthetic 1 are identical with those previously reported.<sup>[1a]</sup> As stated above, our first generation synthesis of 1 was attained in 18 steps from the commercially available 1,4-cyclohexadiene (7).

Studies of the improvements of the first generation asymmetric synthesis: In the first generation asymmetric synthesis described above, some steps (i.e., the conversion of 24 into 26a and the low-yielding transformation of  $35a$  to 1) could be improved by extending our method and searching for more efficient protecting groups and deprotecting conditions.

One-pot conversion of dihydroxy acetal 24 into disilyl aldehydes 26a-e: For the first generation asymmetric synthesis outlined above, the secondary alcohol of 24 was protected as a TBS ether to give 25, and then our deprotection method using acetal groups afforded aldehyde 26 a, both hydroxy groups of which were protected as silyl ethers. Since the second step of the reaction mentioned above, that is, the use of TESOTf/2,4,6-collidine followed by a work-up with water, uses reagents and conditions employed for the silylation of the hydroxy group, we next examined obtaining disilyl

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# Total Synthesis of Scyphostatin **Total Synthesis of Scyphostatin**

aldehyde 26 directly from 24 in one step by applying our method. However, the treatment of 24 with TESOTf/2,4,6 collidine gave the lactol derivative 36, in which the tertiary alcohol was protected as a TES ether, in 63% yield (Scheme 8). We already found that the addition of alcohols to the pyridinium-type salt intermediates prepared by the reaction of the acetal groups with the TESOTf/base combination gave mixed acetals in high yields.<sup>[15b]</sup>



Scheme 8. Treatment of 24 with TESOTf/2,4,6-collidine.

The formation of lactol 36 was rationalized by the intramolecular attack of the naked secondary alcohol on the N,O-acetal carbon atom of the intermediate. This result suggests that the secondary alcohol moiety must be protected before deacetalization. For our deacetalization method, the order of the addition of the reagents and substrates is very important. When the silyl triflate, such as TESOTf or TMSOTf, is added to the mixture of an acetal and 2,4,6-collidine, we can obtain an aldehyde (Method A in Scheme 9).<sup>[15]</sup> On the other hand, when an acetal is added to



Scheme 9. Reactivity of the silyl triflate/2,4,6-collidine combination.

the mixture of the silyl triflate and base, no deacetalization is observed (Method B in Scheme 9). However, the combination of the silyl triflate and 2,4,6-collidine is employed for the silylation of alcohols.[19] We then utilized this different reactivity with 24 for the one-pot synthesis of disilyl aldehyde 26**b** (Scheme 10). Thus 24 was added to a solution of TESOTf (4.0 equiv) and 2,4,6-collidine (8.0 equiv) in  $CH_2Cl_2$ to give disilyl acetal 37 as an intermediate. Additional



Scheme 10. One-pot synthesis of disilyl aldehyde from dihydroxy acetal 24.

Chem. Eur. J. 2007, 13, 10225 – 10238  $\odot$  2007 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim <www.chemeurj.org> – 10229

TESOTf (2.0 equiv) was added to the resulting solution followed by a work-up with water to give the bis(TES) ether aldehyde 26b in 65% yield from a one-pot operation.

The results shown in Scheme 10 encouraged our efforts toward the one-pot synthesis of the disilyl aldehyde, although the yield of 26b was moderate. We then studied this conversion in detail (Table 1). At first, the tert-butyldime-

Table 1. One-pot synthesis of disilyl aldehyde 26.<sup>[a]</sup>

	$R^1$ OTf 2,4,6-collidine (8 equiv) $CH2Cl2$ (0.1 M), temp. 30 min		ЮH HO OMe 24 OMe 5-10 min	$R^2$ OTf 5 min then $H_2O$	$R^3O$ 26	$^{\prime}$ OR $^{\prime}$ CHO
Entry	T	Reagent (equiv)			Product	
	$\lceil$ °C $\rceil$	$\mathbf{R}^1$	$\mathbb{R}^2$		$R^3$	Yield $[\%]$
1	$\mathbf{0}$	TBS(2)	TES $(4)$	26a	<b>TES</b>	85
				26c	<b>TBS</b>	13
2	$\Omega$	TBS(4)	TES (2)	26c	<b>TBS</b>	89
3	$-78$	TBS(2)	TES $(4)$	26 a	<b>TES</b>	98
4	$-78$	TBS(2)	TMS $(4)$	26d	<b>TMS</b>	94
5	$\mathbf{0}$	TIPS $(2)$	TES $(4)$	26 e	<b>TES</b>	90

[a] TIPS = triisopropylsilyl.

thylsilylation of the secondary alcohol of 24 was performed with the proper amount of TBSOTf and an excess of 2,4,6 collidine (8.0 equiv). Thereafter TESOTf or TMSOTf were added to the reaction mixture and a work-up with water afforded disilyl aldehydes 26. The reactions in entries 1 and 2 were carried out at  $0^{\circ}$ C. The use of 2.0 equivalents of TBSOTf as the first silyl triflate and TESOTf as the second silyl triflate gave the desired 26 a along with bis(TBS) ether 26c (Table 1, entry 1). The use of more TBSOTf (4.0 equiv) gave only  $26c$  (Table 1, entry 2). We then studied the reaction using 2.0 equivalents of TBSOTf and 4.0 equivalents of TESOTf at  $-78^{\circ}$ C, which produced of 26a in 98% yield (Table 1, entry 3). The reaction using TMSOTf instead of TESOTf afforded the TMS ether 26 d in 94% yield (Table 1, entry 4). The use of triisopropylsilyl trifluoromethanesulfonate (TIPSOTf) instead of TBSOTf at  $0^{\circ}$ C gave TIPS ether 26e in 90% yield (Table 1, entry 5). Based on these results, we developed a one-pot synthesis of different disilyl aldehydes 26 a–e by controlling the amount of silyl triflate, the order of the addition of the reagents, and the reaction temperature.

Study of the protection and deprotection of the primary alcohol: In the final step of the first generation asymmetric synthesis, the conversion of  $35a$  into 1 resulted in a low yield (32%; Scheme 7). Especially, reagents such as DDQ or CAN,[19] which are recognized as good reagents for the deprotection of MPM ethers, resulted in decomposition of the compound. We therefore examined the protecting group and deprotection conditions for the primary alcohol of 1 using the MPM-type protecting group (Table 2). We used

#### **A EUROPEAN JOURNAL**

Table 2. Examination of protecting group (PG) and deprotection conditions.





[a] NaHCO<sub>3</sub> was added as an additive. [b] Me<sub>2</sub>S was added as an additive. PG=protecting group.

trienamide alcohol 39 as a model compound that has an amino alcohol unit with a highly reactive trienamide component. Treatment of MPM ether 38a with DDO, CAN, or  $MgBr_2 OEt_2^{[20]}$  also resulted in decomposition (Table 2, entries 1–4). However, the use of trityl tetrafluoroborate  $(Ph_3C^+BF_4^{-})^{[18]}$  gave the desired product 39 in 67% yield in 6 hours (Table 2, entry 5). The other trityl reagent, trityl pentafluorophosphate  $(Ph_3C^+PF_5^-)$  produced a trace amount of 39 (Table 2, entry 6), whereas the trityl pentachlorostannane  $(Ph<sub>3</sub>C<sup>+</sup>)$  $SnCl<sub>5</sub><sup>-</sup>$ ) afforded 39 in 56% yield (Table 2, entry 7). The use of the 3,4-dimethoxypheasymmetric synthesis of 1 in a total of 17 steps with a higher total yield via 29b, which has TMS and  $2,4$ DMPM protecting groups. As the silyl group for the protection of the tertiary alcohol, the TMS group was chosen for the higher yield in every



step for the synthesis of 1 (compare Scheme 11 (second generation synthesis) to Scheme 7 (first generation synthesis)). Aldehyde 26d was obtained from commercially available 1,4-cyclohexadiene (7) in eight steps (Schemes 3 and 7 and Table 1, entry 4). Nucleophilic addition of the lithium reagent derived from the 2,4-dimethoxyphenylmethyloxymethyl stannyl compound 40, which was easily prepared by using the procedure developed by Still,<sup>[11]</sup> to the aldehyde  $26d$  afforded alcohol  $(R)$ -27b in 56% yield accompanied by its stereoisomeric alcohol  $(S)$ -27b' in 36% yield. The stereochemistry of the two alcohols was determined by the modified Mosher's method<sup>[12]</sup> (see the Experimental Section). The



Scheme 11. Second generation asymmetric total synthesis of scyphostatin (1). i) <sup>2,4</sup>DMPMOCH<sub>2</sub>SnBu<sub>3</sub> (40);  $nBul.$ ; THF,  $-78^{\circ}$ C (56%); ii) DPPA, PPh<sub>3</sub>, DEAD, THF, RT (75%); iii) LiAlH<sub>4</sub>, THF, 0°C to RT; iv) 30, DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, RT; v) TBAF, THF, RT (59% from azide 28b); vi) TBHP, [VO(acac)<sub>2</sub>], toluene, 0°C  $(73\%)$ ; vii) DMP, CH<sub>2</sub>Cl<sub>2</sub>, 40<sup>o</sup>C (69%, 78% based on the consumed 33b); viii) Ph(Cl)S=NtBu, LDA, [15]crown-5, THF,  $-78$ °C (35%, 82% based on the consumed 34b); ix)  $Ph_3C^{+}BF_4^-$ ,  $CH_2Cl_2$ , 0°C (66%).

nylmethyl  $(^{3,4}$ DMPM) group and Ph<sub>3</sub>C<sup>+</sup>BF<sub>4</sub><sup>-</sup> increased the yield to 90% and shortened the reaction time to 2 hours (Table 2, entry 8). The use of the 2,4-dimethoxyphenylmethyl  $(^{2,4}$ DMPM) group and Ph<sub>3</sub>C<sup>+</sup>BF<sub>4</sub><sup>-</sup> gave better results:  $<$  5 minutes and 93% yield of 39 (Table 2, entry 9). Although  $HBF_4$  also afforded 39 in a quantitative yield, the conditions were strongly acidic (Table 2, entry 10). We then chose <sup>2,4</sup>DMPM and  $Ph_3C^+BF_4^-$  for the most efficient total synthesis of 1.

Asymmetric total synthesis of 1: second generation: Since the improvement of some of the drawbacks in the first generation asymmetric synthesis of 1 was carried out, we tried to utilize these results and achieved the second generation

Mitsunobu reaction of 27b with DPPA<sup>[13]</sup> afforded azide(S)-**28b** in 75% yield, which was reduced by  $LiAlH<sub>4</sub>$  to afford amino alcohol 29b. Condensation of the amino group of **29b** with the side chain acid  $30$ ,<sup>[5a]</sup> followed by desilylation gave dihydroxyamide 32 b in 59% yield from azide 28 b. The stereo- and chemoselective epoxidation of 32b with [VO- $(\text{acac})_2$ ] and TBHP<sup>[14]</sup> gave epoxy alcohol **33b** in 73% yield, which was oxidized using Dess–Martin periodinane  $(DMP)^{[21]}$  to afford epoxy ketone 34b in 69% yield (78%) based on consumed 33b). The treatment of 34b with LDA and reaction with  $N$ -tert-butylphenylsulfinimidoyl chloride<sup>[17]</sup> gave enone 35b in 35% yield (82% based on consumed 34b). Finally, deprotection of  $^{2,4}$ DMPM ether 35b was successfully carried out by  $Ph_3C^+BF_4^{-[18]}$  to afford 1 in 66%

# Total Synthesis of Scyphostatin **FULL PAPER**

yield. The spectroscopic data of synthetic 1 were identical with those of the authentic compound.

#### Conclusion

The concise asymmetric total synthesis of scyphostatin (1) was achieved from commercially available 1,4-cyclohexadiene in a total of 17 steps. The characteristic points of our synthesis are that 1) the chiral acetal from the optically pure hydrobenzoin works not only as a discrimination tool for two olefins, a source of oxygen atoms, and a protecting group for the alcohol unit, but also as the template for stereoselective oxidation using  $\text{SeO}_2$ <sup>[8]</sup> 2) the silyl triflate/2,4,6collidine combination works not only for the deprotection of the acetal, but also for the silylation of hydroxy functions to give disilyl aldehyde compounds with different kinds of silyl groups in a one-pot operation; 3) the use of the <sup>2,4</sup>DMPM protecting group would be very useful for the syntheses of other unstable compounds; and 4) the asymmetric centers are constructed step-by-step, which allows the syntheses of many stereoisomeric analogues.

### Experimental Section

General: All melting points are uncorrected. <sup>1</sup>H NMR spectra were recorded at 270 or 300 MHz and 13C NMR spectra were recorded at 67.8 or 75 MHz with CDCl<sub>3</sub> as the solvent and  $\text{SiMe}_4$  as the internal standard. Infrared (IR) absorption spectra  $(cm<sup>-1</sup>)$  were recorded as KBr pellets.

1-(2,2-Diethoxyethyl)-2,5-cyclohexadiene (8): sec-BuLi (0.98m in nhexane/cyclohexane, 3.2 mL, 3.1 mmol) was added slowly to a stirred solution of 7 (0.30 mL, 3.2 mmol) in THF (5 mL) at  $-78$ °C under N<sub>2</sub>. After being stirred for 5 min at the same temperature, TMEDA (0.50 mL, 3.3 mmol) was added. The resulting mixture was stirred for an additional 1 h. Bromoacetoaldehyde diethyl acetal (0.40 mL, 2.7 mmol) was added to the reaction mixture. After being stirred for 2 h, the reaction mixture was quenched by addition of saturated NaHCO<sub>3</sub>. The reaction mixture was extracted with Et<sub>2</sub>O. The organic layer was washed with brine, dried over  $MgSO<sub>4</sub>$ , and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/Et<sub>2</sub>O (10:1) as the eluent to give  $\frac{8}{10}$  (0.39 g, 2.0 mmol, 75%) as a colorless oil. <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3): \delta = 1.17 - 1.25 \text{ (6H, m)}, 1.72 \text{ (2H, t, } J = 5.7 \text{ Hz}), 2.61 -$ 2.66 (2H, m), 2.83–2.94 (1H, m), 3.45–3.71 (4H, m), 4.65 (1H, t, J= 5.7 Hz), 5.65–5.74 ppm (4H, m).

(4R,5R)-2-(2,5-Cyclohexadienylmethyl)-4,5-diphenyl-1,3-dioxolane (2): p-TsOH (0.1 mmol) was added slowly to a stirred solution of 8 (0.31 g, 1.6 mmol) and  $(R,R)$ -hydrobenzoin  $(0.34 \text{ g}, 1.6 \text{ mmol})$  in toluene  $(10 \text{ mL})$ under  $N_2$ . The resulting mixture was stirred for 2 h at 70 °C. After being cooled to room temperature, saturated  $NaHCO<sub>3</sub>$  (aq.) was added to the mixture. The reaction mixture was extracted with EtOAc. The organic layer was washed with brine, dried over  $MgSO<sub>4</sub>$ , and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc  $(25:1)$  as the eluent to give 2  $(0.51 \text{ g}, 1.6 \text{ mmol})$ , 100%) as a colorless oil.  $[\alpha]_D^{25} = +31.0$  (c=1.00, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} =$ 3022, 1605, 1497, 1456 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.02 (2H, dd, J=4.8, 6.7 Hz), 2.6–2.7 (2H, m), 3.1–3.2 (1H, m), 4.77 (2H, s), 5.63 (1H, t, J=4.8 Hz), 5.7–5.9(4H, m), 7.2–7.4 ppm (10H, m); elemental analysis (%) calcd for  $C_{22}H_{22}O_2$ : C 82.99, H 6.96; found: C 82.93, H 7.04. (1R,3R,4R,6R,8R,12R)-12-Bromo-6-methoxy-3,4-diphenyl-2,5-

dioxabicyclo[6.4.0]dodec-9-ene (3): NBS (0.21 g, 1.2 mmol) was added portionwise to a stirred solution of 2 (0.32 g, 1.0 mmol) and MeOH (0.20 mL, 5.0 mmol) in CH<sub>3</sub>CN (10 mL) at  $-40^{\circ}$ C under N<sub>2</sub>. The resulting mixture was allowed to warm to room temperature over 6 h with stirring. After saturated NaHCO<sub>3</sub> (aq.) was added, The reaction mixture was extracted with  $Et<sub>2</sub>O$ . The organic layer was washed with brine, dried over MgSO4, and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc (15:1) as the eluent to give 3 (0.27 g, 0.63 mmol, 64%) as a colorless oil.  $\lbrack a \rbrack_{D}^{24} = -156.0$  ( $c = 1.62$ , CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3031, 1495, 1454, 1125$  cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.01 (2H, dd, J = 5.6, 6.3 Hz), 2.5–2.6 (1H, m), 2.8–3.0 (1H, m), 3.29(3H, s), 3.3–3.5 (1H, m), 4.04 (1H, dd, J=5.3, 8.9Hz), 4.31 (1H, dt, J=7.9, 5.9 Hz), 4.41 (1H, d, J=9.1 Hz), 4.47 (1H, d, J=9.1 Hz), 5.24  $(1\,\text{H}, \text{dd}, J = 5.0, 6.3 \text{ Hz}), 5.5 - 5.7 (2\,\text{H}, \text{m}), 6.9 - 7.4 \text{ ppm} (10\,\text{H}, \text{m});$  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 35.3, 35.4, 37.7, 47.7, 54.9, 81.9, 88.0,$ 88.5, 105.1, 124.4, 127.3, 127.4, 127.5, 127.7, 127.8, 129.1, 137.9, 138.1 ppm; FAB-MS  $m/z$ : 451  $[M^+ + Na]$ ; HR-FAB-MS: calcd for  $C_{23}H_{25}O_3BrNa$  [ $M^+ +Na$ ]: 451.0885; found 451.0873.

(1R,3R,4R,6R,8S)-6-Methoxy-3,4-diphenyl-2,5-dioxabicyclo[6.4.0]dodec-**9-ene** (9): A mixture of 3 (0.91 g, 2.1 mmol),  $Bu_3SnH$  (0.86 mL, 3.2 mmol), and a catalytic amount of azobisisobutyronitrile (AIBN) in toluene (21 mL) was refluxed for 1 h under  $N_2$ . After being cooled to room temperature, saturated KF (aq.) was added to the mixture. The reaction mixture was extracted with EtOAc. The organic layer was washed with brine, dried over  $MgSO_4$ , and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/ EtOAc  $(10:1)$  as the eluent to give 9  $(0.66 \text{ g}, 1.9 \text{ mmol}, 89\%)$  as colorless crystals. M.p. 162–164 °C (hexane/EtOAc);  $[a]_D^{26} = -101.2$  (c=1.23, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3029, 1096, 994 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$ =1.7–2.2 (6H, m), 3.2–3.3 (1H, m), 3.29 (3H, s), 3.84 (1H, dt, J=4.5, 11.4 Hz), 4.42 (1H, d, J=9.0 Hz), 4.49 (1H, d, J=9.0 Hz), 5.39 (1H, dd,  $J=5.1, 8.1$  Hz), 5.63 (2H, d,  $J=2.7$  Hz), 6.9–7.3 ppm (10H, m); <sup>13</sup>C NMR  $(75 \text{ MHz}, \text{CDCl}_3)$ :  $\delta = 24.6$ , 25.7, 35.4, 37.3, 54.8, 79.6, 87.4, 88.7, 105.3, 126.9, 127.3, 127.6, 127.7, 127.9, 129.1, 138.3, 139.5 ppm; elemental analysis (%) calcd for C<sub>23</sub>H<sub>26</sub>O<sub>3</sub>: C 78.83, H 7.48; found: C 78.86, H 7.47.

(1S,3R,5R,6R,8S)-3-Methoxy-5,6-diphenyl-4,7-dioxabicyclo[6.4.0]dodec-11-en-1-ol (4): A mixture of 9 (71 mg, 0.202 mmol), pyridine (32.6  $\mu$ L, 0.404 mmol), and  $SeO<sub>2</sub>$  (11.4 mg, 0.102 mmol) in 1,4-dioxane (4 mL) was stirred for 1 h at 80 $^{\circ}$ C under N<sub>2</sub>. After being cooled to room temperature, saturated NaHCO<sub>3</sub> (aq.) was added to the mixture. The reaction mixture was extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc (7:1) as the eluent to give 4 (30.6 mg, 0.084 mmol, 41%) and recovered 9 (20.6 mg, 29%). 4: White crystals; M.p. 159–162 °C (hexane);  $[a]_D^{26}$  =  $-34.3$  (c=1.21, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3438$ , 3033, 1096, 994 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.7–1.8 (1H, m), 1.9–2.0 (1H, m), 2.0– 2.2 (2H, m), 2.07 (1H, dd,  $J=7.2$ , 15.2 Hz), 2.44 (1H, dd,  $J=4.5$ , 15.2 Hz), 3.32 (3H, s), 4.08 (1H, dd, J=3.3, 10.8 Hz), 4.51 (1H, d, J= 9.0 Hz), 5.01 (1H, d, J=9.0 Hz), 5.47 (1H, m), 5.50 (1H, s), 5.73 (1H, dt,  $J=3.5, 9.9$  Hz), 6.9–7.3 ppm (10H, m); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 24.2, 27.3, 41.5, 55.0, 75.7, 84.0, 87.2, 89.0, 106.5, 127.1, 127.5, 127.6, 127.7, 127.9, 128.0, 128.1, 132.5, 137.3, 139.5 ppm; elemental analysis calcd (%) for C<sub>23</sub>H<sub>26</sub>O<sub>4</sub>: C 75.38, H 7.15; found: C 75.43, H 7.12.

 $(1S,6S)$ -1-(2,2-Dimethoxyethyl)-6- $[(1R,2R)$ -2-hydroxyethoxy-1,2-diphenyl]cyclohex-2-en-1-ol (10): PPTS (10 mg, 0.04 mmol) was added to a stirred solution of 4 (140 mg, 0.39 mmol) in MeOH (4 mL) at room temperature under  $N_2$ . The reaction mixture was stirred at room temperature overnight. The reaction mixture was quenched with saturated  $NAHCO<sub>3</sub>$ (aq.) and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO<sub>4</sub>, and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc (2:1) as the eluent to give 10 (140 mg, 0.35 mmol, 91%) as colorless crystals. M.p. 82–84 °C (hexane);  $\left[\alpha\right]_D^{25} = -14.4$  (c=1.19, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} =$ 3455, 3031, 1455, 1240 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.4–1.7  $(2H, m)$ , 1.90 (1H, dd,  $J=3.7$ , 14.5 Hz), 1.9–2.1 (2H, m), 2.49 (1H, dd,  $J=8.6, 14.5 \text{ Hz}$ ), 3.41 (3H, s), 3.47 (3H, s), 3.80 (1H, dd,  $J=4.1$ , 11.9 Hz), 4.1-4.4 (1H, brs), 4.41 (1H, d,  $J=7.2$  Hz), 4.68 (1H, d,  $J=$ 7.2 Hz), 4.7–5.0 (1H, brs), 4.82 (1H, dd, J = 3.7, 8.6 Hz), 6.9–7.3 ppm (10H, m); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 24.7, 25.8, 37.9, 52.2, 53.9, 73.5, 80.1, 84.0, 89.3, 102.7, 127.0, 127.4, 127.7, 131.6, 139.6, 139.8 ppm; el-

#### **A EUROPEAN JOURNAL**

emental analysis (%) calcd for  $C<sub>24</sub>H<sub>30</sub>O<sub>5</sub>$ : C 72.34, H 7.59; found: C 72.39, H 7.61.

(1S,2S)-2-(2,2-Dimethoxyethyl)cyclohex-3-ene-1,2-diol (24): Ca (49.3 mg, 1.23 mmol) was added to liquid NH<sub>3</sub> (3.0 mL) at  $-78^{\circ}$ C under N<sub>2</sub>. The reaction mixture was allowed to warm to  $-40^{\circ}$ C to give a clear-blue solution. A solution of 10 (49.0 mg, 0.123 mmol) and EtOH (0.06 mL, 1.23 mmol) in  $Et_2O$  (0.5 mL) was added to dropwise to the blue solution at  $-40$ °C under N<sub>2</sub>. After being stirred for 30 min, the reaction mixture was quenched with saturated  $NH<sub>4</sub>Cl$  (aq.) After  $NH<sub>3</sub>$  was removed by distillation at atmospheric pressure, The reaction mixture was extracted with EtOAc. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc (1:2) as the eluent to give 24 (22.2 mg, 0.110 mmol, 91%) as a colorless oil.  $[\alpha]_D^{28} = -28.3$  (c=2.10, CHCl<sub>3</sub>): IR (KBr):  $\tilde{v} = 3340, 1724, 1380, 1274 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta =$ 1.66–1.83 (2H, m), 1.92–2.04 (1H, m), 2.13–2.21 (3H, m), 3.36 (3H, s), 3.44 (3H, s), 3.54 (1H, s), 3.81 (1H, dd, J=3.9, 7.5 Hz), 3.85 (1H, s), 4.76  $(1\,\text{H}, \text{dd}, J = 3.9, 7.5 \text{ Hz})$ , 5.51  $(1\,\text{H}, \text{dt}, J = 10.0, 2.1 \text{ Hz})$ , 5.74 ppm  $(1\,\text{H}, \text{dt},$  $J=10.0$ , 3.6 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 22.7, 25.4, 38.2, 51.8$ 53.6, 71.5, 72.4, 102.1, 127.9, 130.8 ppm; FAB-MS m/z: 225 [M++Na]; HR-FAB-MS: calcd for  $C_{10}H_{18}O_4$ Na  $[M^+ + Na]: 225.1103$ ; found 225.1095.

#### (1S,6S)-1-(2,2-Dimethoxyethyl)-6-[(1,1-dimethylethyl)(dimethyl)siloxy]-

cyclohex-2-en-1-ol (25): Imidazole (71.6 mg, 1.06 mmol) and TBSCl (158.6 mg, 1.06 mmol) were added successively to a solution of 24 (30.4 mg, 0.154 mmol) in DMF (1.5 mL) at room temperature under  $N_2$ , and the mixture was stirred for 12 h. The reaction mixture was quenched with addition of water and extracted with EtOAc. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc  $(7:1)$  as the eluent to give 25 (46 mg, 0.14 mmol, 98%) as a colorless oil.  $[\alpha]_D^{25} = -15.5$  (c=1.29, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3487$ , 2952, 1471, 1253 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.10 (6H, s), 0.91 (9H, s), 1.56–1.65 (1H, m), 1.77–1.84 (2H, m), 2.09–2.16 (3H, m), 3.35 (3H, s), 3.38 (3H, s), 3.64 (1H, s), 3.77 (1H, dd,  $J=3.3$ , 9.6 Hz), 4.78 (1H, dd,  $J=$ 5.2, 6.8 Hz), 5.52 (1H, dt,  $J=10.0$ , 1.8 Hz), 5.67 ppm (1H, dt,  $J=10.0$ , 3.3 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = -4.8, 18.0, 23.5, 27.6, 37.7, 52.2, 53.0, 71.9, 74.7, 102.3, 127.6, 131.2 ppm; elemental analysis (%) calcd for  $C_{16}H_{32}O_4Si$ : C 60.72, H 10.19; found: C 60.64, H 10.00.

(1S,6S)-1-Triethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2-en-1-yl acetaldehyde (26 a): 2,4,6-Collidine (62  $\mu$ L, 0.472 mmol) and TESOTf (71  $\mu$ L, 0.315 mmol) were successively added to a solution of 25 (24.9mg, 0.0787 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (0.78 mL) at 0 °C under N<sub>2</sub>. After being stirred at the same temperature for 5 min, the reaction mixture was quenched with water and extracted with  $CH<sub>2</sub>Cl<sub>2</sub>$ . The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated in vacuo. The residue was purified by column chromatography using hexane/EtOAc  $(25:1)$  as the eluent to give 26 a  $(28.1 \text{ mg}, 0.073 \text{ mmol}, 93\%)$  as a colorless oil.  $\left[\alpha\right]_D^{25} = +36.9$  (c=1.60, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 2954$ , 1720, 1461, 1253 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.08 (6H, s), 0.62 (6H, q, J = 7.8 Hz), 0.88 (9H, s), 0.94 (9H, t, J=7.8 Hz), 1.54–1.64 (1H, m), 1.82– 1.92 (1H, m), 2.03–2.19 (2H, m), 2.31 (1H, dd, J=2.7, 15.9Hz), 2.63 (1H, dd, J=3.2, 15.9Hz), 3.89 (1H, dd, J=2.0, 14.5 Hz), 5.57 (1H, dt,  $J=10.0$ , 1.8 Hz), 5.76 (1H, dt,  $J=10.0$ , 3.6 Hz), 9.82 ppm (1H, dd,  $J=2.7$ , 3.2 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = -4.0, 7.2, 18.0, 23.1, 25.9, 28.2, 51.4, 75.8, 129.5, 130.9, 202.6; elemental analysis (%) calcd for  $C_{20}H_{40}O_3Si_2$ : C 62.44, H 10.48; found: C 62.55, H 10.38 ppm.

Preparation of MPMOCH<sub>2</sub>SnBu<sub>3</sub> (13):<sup>[11]</sup> A mixture of 4-methoxybenzyl alcohol (3.8 mL, 30.4 mmol) and NaH (1.4 g, 60% in oil, 35.1 mmol) in THF (120 mL) was stirred at room temperature under  $N_2$ . After being stirred for 1 h, a solution of tri-n-butylstannylmethyl iodide (10.1 g, 23.4 mL) in THF (30 mL) was added slowly to the solution. The resulting mixture was stirred for three days. The reaction mixture was quenched by addition of MeOH. The solution was diluted with  $Et<sub>2</sub>O$ . The organic layer was washed with water two times. The organic layer was dried over Na2SO4 and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/Et<sub>2</sub>O (30:1) as the eluent to give 13 (8.8 g, 85%) as a colorless oil. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  =

0.86–0.94 (15H, m), 1.29(6H, m), 1.49(6H, m), 3.71 (2H, s), 3.78 (3H, s), 4.34 (2H, s), 6.86 (2H, d,  $J=8.7$  Hz), 7.22 ppm (2H, d,  $J=8.7$  Hz);  $^{13}$ C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 8.9, 13.7, 27.3, 29.1, 55.1, 61.0, 77.4,$ 113.5, 129.0, 130.9, 158.9 ppm.

(2R)-1-[(1S,6S)-1-Triethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2-en-1-yl]-3-[(4-methoxyphenyl)methyloxy]propan-2-ol (27 a) and (2S)-1- [(1S,6S)-1-triethylsiloxy-6-tert-butyldimethylsiloxy cyclohex-2-en-1-yl]-3- [(4-methoxyphenyl)methyloxy]propan-2-ol (27 a'): nBuLi (1.56m in hexane,  $0.22$  mL) was added dropwise to a solution of MPMOCH<sub>2</sub>SnBu<sub>3</sub> (13; 155.4 mg, 0.352 mmol) in dry THF (1.0 mL) at  $-78^{\circ}$ C under N<sub>2</sub>. After being stirred for 15 min, a solution of 26 a (33.8 mg, 0.088 mmol) in THF (1.0 mL) was added to the reaction mixture. After being stirred for 5 min, the reaction mixture was quenched with water and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated in vacuo. The residue was purified by column chromatography using hexane/EtOAc  $(7:1)$  as the eluent to give 27a  $(26.4 \text{ mg})$ , 56%) and 27 a' (12 mg, 25%).

**27a**: Colorless oil;  $[a]_D^{25} = +33.0$  ( $c = 1.25$ , CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3482$ , 2945, 1612, 1514, 1461 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.08 (3 H, s), 0.09 (3H, s), 0.62 (6H, q,  $J=7.8$  Hz), 0.83 (9H, s), 0.94 (9H, t,  $J=$ 7.8 Hz), 1.58–2.19(6H, m), 3.30–3.40 (2H, m), 3.80 (3H, s), 3.97 (1H, d,  $J=5.4$  Hz), 4.05 (1H, s), 4.11-4.20 (1H, m), 4.46 (1H, d,  $J=11.7$  Hz), 4.53 (1 H, d,  $J=11.7$  Hz), 5.53 (1 H, d,  $J=10.2$  Hz), 5.78 (1 H, dt,  $J=10.2$ , 3.6 Hz), 6.86 (2H, d,  $J=8.4$  Hz), 7.26 ppm (2H, d,  $J=8.4$  Hz); <sup>13</sup>C NMR  $(75 \text{ MHz}, \text{CDCl}_3)$ :  $\delta = -4.9, -4.0, 6.6, 7.0, 17.9, 22.2, 25.7, 26.3, 40.8, 55.1,$ 67.3, 72.8, 74.5, 76.2, 113.5, 129.1, 129.8, 130.5, 131.0, 159.0 pm; elemental analysis (%) calcd for  $C_{29}H_{52}O_5Si_2$ : C 64.88, H 9.76; found: C 64.74, H 9.79.

**27a'**: Colorless oil;  $[a]_D^{26} = -12.5$  (c=1.39, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3482$ , 2931, 1612, 1514, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.11 (3 H, s), 0.14 (3H, s), 0.62 (6H, q,  $J=7.2$  Hz), 0.91 (9H, s), 0.93 (9H, t,  $J=$ 7.2 Hz), 1.62–1.77 (3H, m), 1.98 (1H, dd, J=9.0, 14.7 Hz), 2.07–2.13 (2H, m), 3.35–3.46 (2H, m), 3.80 (3H, s), 3.84 (1H, dd, J=4.2, 10.8 Hz), 4.11– 4.18 (1H, m), 4.18 (1H, s), 4.50 (2H, s), 5.62–5.64 (2H, m), 6.86 (2H, d,  $J=8.7$  Hz), 7.26 ppm (2H, d,  $J=8.7$  Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = -4.6, -4.4, 6.6, 6.8, 7.0, 7.1, 18.0, 24.5, 25.7, 25.8, 28.8, 41.8, 55.1, 66.0, 72.7, 74.5, 77.0, 113.5, 127.7, 129.1, 130.6, 133.0, 158.9 ppm; FAB-MS m/z: 559  $[M^+ + Na]$ ; HR-FAB-MS: calcd for C<sub>29</sub>H<sub>52</sub>O<sub>5</sub>Si<sub>2</sub>Na  $[M^+ + Na]$ : 559.3251; found 559.3265.

Determination of the absolute configurations of the secondary alcohols of 27 a and 27 a': The absolute configurations of the secondary alcohols of 27a and 27a' were determined by a modified Mosher's method.<sup>[12]</sup>



General procedure for the preparation of an MTPA ester: DCC (3.3 mmol), a-methoxy-a-(trifluoromethyl)phenylacetyl chloride (MTPACl) (4.3 mmol), and DMAP (3.3 mmol) were added successively to a stirred solution of secondary alcohol  $27a$  or  $27a'$  (1.0 mmol) in

# Total Synthesis of Scyphostatin **Total Synthesis of Scyphostatin**

 $CH_2Cl_2$  (10.0 mL) at room temperature. After being stirred for 30 min, the solvent was removed in vacuo. The residue was purified by column chromatography using hexane/EtOAc (20:1) as the eluent.

(2R)-1-[(1S,6S)-1-Triethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2-en-



1-yl]-3-[(4-methoxyphenyl)methyloxy]propan-2 yl (2S)-3,3,3-trifluoro-2-methoxy-2-phenylpropanoate  $(27a; (S)$ -MTPA): Colorless oil; <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3): \delta = 0.07 \text{ (6H, s)}, 0.62 \text{ (6H, q)}$ J=7.8 Hz), 0.87 (9H, s), 0.94 (9H, t, J=7.8 Hz), 1.37–2.08 (5H, m), 2.15 (1H, dd,  $J=4.8$ , 15.0 Hz), 3.44 (1 H, d,  $J=10.8$  Hz), 3.49 (3 H, s), 3.56 (1H, dd,  $J=2.1$ , 10.8 Hz), 3.76 (1H, dd,  $J=$ 3.6, 11.1 Hz), 3.80 (3H, s), 3.99–4.07 (1H, m), 4.32 (1H, d,  $J=11.4$  Hz), 4.41 (1H, d,  $J=$ 

11.4 Hz), 5.62 (1H, d,  $J=10.0$  Hz), 5.69 (1H, d,  $J=10.0$  Hz), 6.83 (2H, d,  $J=8.4$  Hz), 7.16 (2H, d,  $J=8.4$  Hz), 7.27–7.35 (3H, m), 7.57 ppm (2H, d,  $J=7.5$  Hz).

(2R)-1-{(1S,6S)-1-Triethylsiloxy-6-[tert-butyldimethylsiloxycyclohex-2-en-1-yl]}-3-[(4-methoxyphenyl)methyloxy]propan-2-yl (2R)-3,3,3-trifluoro-2 methoxy-2-phenylpropanoate (27a;  $(R)$ -MTPA): Colorless oil; <sup>1</sup>H NMR



(300 MHz, CDCl<sub>3</sub>):  $\delta = 0.05$  (6H, s), 0.61 (6H, q,  $J=7.8$  Hz), 0.85 (9H, s), 0.94 (9H, t,  $J=7.8$  Hz), 1.42–1.72 (3H, m), 1.96–2.00 (2H, m), 2.11 (1H, dd,  $J=5.1$ , 15.3 Hz), 3.52 (1H, d,  $J=10.8$  Hz), 3.56 (3H, s), 3.65 (1H, dd, J=2.7, 11.0 Hz), 3.72  $(1\,\text{H}, \text{dd}, J=3.6, 11.0 \text{ Hz})$ , 3.81  $(3\,\text{H}, \text{s})$ , 4.41  $(1\,\text{H},$ d,  $J=11.4$  Hz), 4.53 (1H, d,  $J=11.4$  Hz), 5.44  $(1\,\text{H}, \text{dt}, J=10.2, 3.0 \text{ Hz}), 5.58 (1\,\text{H}, \text{d}, J=$ 10.2 Hz), 5.63–5.73 (1H, m), 6.85 (2H, d,  $J=$ 8.7 Hz), 7.23 (2H, d, J=8.7 Hz), 7.28–7.37 (3H, m), 7.59 ppm (2H, d,  $J=7.5$  Hz).

(2S)-1-{(1S,6S)-1-Triethylsiloxy-6-[tert-butyldimethylsiloxycyclohex-2-en-1-yl]}-3-[(4-methoxyphenyl)methyloxy]propan-2-yl (2S)-3,3,3-trifluoro-2-



methoxy-2-phenylpropanoate (27 a'; (S)-MTPA): Colorless oil; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.04 (3H, s), 0.07 (3H, s), 0.57 (6H, q,  $J=$ 7.8 Hz), 0.85 (9H, s), 0.91 (9H, t, J=7.8 Hz), 1.50–1.98 (5H, m), 2.03 (1H, dd,  $J=4.5$ , 10.5 Hz), 3.52–3.54 (1H, m), 3.55 (3H, s), 3.70  $(1\,\text{H}, \text{dd}, J=3.6, 9.6 \text{ Hz}), 3.79 \text{ (1 H}, \text{dd}, J=2.4,$ 11.4 Hz), 3.80 (3H, s), 4.38 (1H, d,  $J=11.4$  Hz), 4.52 (1H, d, J=11.4 Hz), 5.40 (1H, d, J= 10.0 Hz), 5.61 (1H, dt,  $J=10.0$ , 3.6 Hz), 5.70– 5.75 (1H, m), 6.84 (2H, d, J=8.7 Hz), 7.22 (2H, d, J=8.7 Hz), 7.27–7.41

 $(3H, m)$ , 7.59 ppm  $(2H, d, J=7.2 \text{ Hz})$ .

(2S)-1-{(1S,6S)-1-Triethylsiloxy-6-[tert-butyldimethylsiloxycyclohex-2-en-1-yl]}-3-[(4-methoxyphenyl)methyloxy]propan-2-yl (2R)-3,3,3-trifluoro-2 methoxy-2-phenylpropanoate (27 a'; (R)-MTPA):



Colorless oil; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.07 (3H, s), 0.09 (3H, s), 0.58 (6H, q,  $J=$ 7.5 Hz), 0.86 (9H, s), 0.91 (9H, t, J=7.5 Hz), 1.53–1.94 (5H, m), 2.09 (1H, dd,  $J=4.2$ , 14.4 Hz), 3.45 (1 H, dd,  $J=8.1$ , 11.4 Hz), 3.51 (3H, s), 3.72 (1H, dd, J=2.4, 11.4 Hz), 3.78 (3H, s), 3.84 (1 H, dd,  $J=3.3$ , 9.6 Hz), 4.31 (1 H, d,  $J=$ 11.7 Hz), 4.42 (1 H, d,  $J=11.7$  Hz), 5.49 (1 H, d,  $J=10.2$  Hz), 5.67 (1H, dt,  $J=10.2$ , 3.9 Hz), 5.71–

5.77 (1H, m), 6.82 (2H, d, J=8.7 Hz), 7.16 (2H, d, J=8.7 Hz), 7.25–7.42  $(3H, m)$ , 7.56 ppm  $(2H, d, J=7.5 Hz)$ .

(2S)-2-Azido-1-{(1S,6S)-1-triethylsiloxy-6-[tert-butyldimethylsiloxycyclohex-2-en-1-yl]}-3-[(4-methoxyphenyl)methyloxy]propane (28 a): PPh<sub>3</sub> (194.0 mg, 0.738 mmol), DEAD (0.33 mL; 40% in toluene, 0.738 mmol), and DPPA  $(95 \mu L, 0.442 \text{ mmol})$  were added successively to a stirred solution of  $27a$  (207.4 mg, 0.369 mmol) in THF (3.7 mL) at room temperature under Ar. The reaction mixture was stirred at the same temperature for 30 min. After removal of the solvent in vacuo, the residue was purified by column chromatography using hexane/EtOAc (20:1) as the eluent to give **28 a** (138.8 mg, 0.247 mmol, 67%) as a colorless oil.  $[a]_D^{25} = +19.7$ 

 $(c=0.77, \text{ CHCl}_3); \text{ IR } (KBr): \tilde{\nu}=2952, 2104, 1612, 1514, 1461 \text{ cm}^{-1};$ <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.08 (3H, s), 0.11 (3H, s), 0.58 (6H, q,  $J=7.8$  Hz), 0.88 (9H, s), 0.92 (9H, t,  $J=7.8$  Hz), 1.50–1.65 (3H, m), 1.84– 1.92 (3H, m), 1.98–2.12 (1H, m), 3.41 (1H, dd, J=8.4, 9.9 Hz), 3.66 (1H, dd,  $J=3.3$ , 9.9 Hz), 3.81 (3H, s), 3.86 (1H, dd,  $J=2.8$ , 8.4 Hz), 4.42 (2H, s), 5.48 (1H, d,  $J=10.1$  Hz), 5.74 (1H, dt,  $J=10.1$ , 3.7 Hz), 6.88 (2H, d,  $J=8.6$  Hz), 7.28 ppm (2H, d,  $J=8.6$  Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = -4.1, -4.0, 6.9, 7.0, 14.1, 18.0, 22.5, 25.8, 27.2, 38.9, 55.1, 58.2, 72.7,$ 73.6, 74.0, 74.5, 113.6, 113.7, 128.9, 129.0, 129.2, 129.6, 130.1, 131.3, 159.1 ppm; FAB-MS  $m/z$ : 584 (M<sup>+</sup>+Na); HR-FAB-MS: calcd for  $C_{29}H_{51}N_3O_4Si_2Na$  [ $M^+$ +Na]: 584.3316; found 584.3322.

### (1S,6S)-1-{(2S)-3-[(4-methoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)-

(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}-2-cyclohexene-1,6-diol (32a): A solution of 28a (138.8 mg, 0.247 mmol) in dry THF (2.4 mL) was added to a solution of  $LiAlH_4$  $(93.7 \text{ m} \sigma, 2.47 \text{ mmol})$  in THF  $(2.4 \text{ mL})$  at  $0^{\circ}$ C under Ar. After being stirred for 1.5 h at room temperature, the reaction mixture was quenched with water and 15% NaOH (aq.), then the precipitate was filtered through a Celite pad. The filtrate was evaporated in vacuo to give crude 29 a. Side-chain carboxylic acid 30 (86.0 mg, 0.296 mmol), DCC (203.9 mg, 0.988 mmol), and DMAP (120.7 mg, 0.988 mmol) were added to a solution of crude  $29a$  in CH<sub>2</sub>Cl<sub>2</sub> (2.5 mL) at room temperature under Ar. After being stirred for 1 h, the reaction mixture was quenched with water and extracted with  $CH_2Cl_2$ . The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo to give crude coupled product 31 a. TBAF (2.47 mL, 1.0m in THF, 2.47 mmol) was added to crude  $31a$  in THF (1.2 mL) at room temperature under Ar. After being stirred at room temperature for 10 h, the reaction mixture was quenched with water and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using hexane/EtOAc (1:2) as the eluent to give **32a** (97.6 mg, 66%) as a yellow oil.  $[a]_D^{24} =$  $+3.13$  (c=1.53, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3307$ , 2923, 1651, 1608, 1514, 1456 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.79 (3H, d, J = 6.4 Hz), 0.84  $(3H, t, J=7.3 Hz)$ , 0.90  $(3H, d, J=6.6 Hz)$ , 0.99  $(3H, d, J=6.6 Hz)$ , 1.03 (1H, m), 1.16–1.34 (3H, m), 1.52 (3H, d, J=1.3 Hz), 1.55 (1H, m), 1.72– 1.80 (2H, m), 1.83–1.92 (3H, m), 1.98–2.09 (2H, m), 2.17 (1H, m), 2.23 (1H, m), 2.32 (1H, m), 3.60 (1H, d, J=4.2 Hz), 3.81 (3H, s), 3.92 (1H, dd, J=2.9, 8.3 Hz), 4.26 (1H, m), 4.42 (1H, d, J=11.5 Hz), 4.50 (1H, d,  $J=11.5$  Hz), 4.84 (1H, d,  $J=9.4$  Hz), 5.51 (1H, d,  $J=10.0$  Hz), 5.67-5.72  $(2H, m)$ , 5.76 (1H, d, J=14.9 Hz), 6.07 (1H, dd, J=10.9, 14.9 Hz), 6.16 (1H, dd,  $J=11.4$ , 14.7 Hz), 6.27 (1H, d,  $J=6.0$  Hz), 6.48 (1H, dd,  $J=$ 10.9, 14.7 Hz), 6.89 (2 H, d,  $J=9.0$  Hz), 7.22 (1 H, dd,  $J=11.4$ , 14.9 Hz), 7.26 ppm (2H, d, J=9.0 Hz); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>);  $\delta$  = 12.1, 16.2. 19.5, 21.1, 21.4, 22.9, 25.7, 28.3, 30.6, 34.1, 34.9, 39.1, 44.0, 45.9, 48.3, 55.3, 72.0, 72.3, 72.6, 72.9, 113.9, 121.9, 127.7, 128.3, 129.2, 129.6, 129.8, 131.3, 132.1, 133.1, 140.7, 141.9, 145.7, 159.4, 166.9 ppm; FAB-MS m/z: 594 [M<sup>+</sup> +H]; HR-FAB-MS: calcd for  $C_{37}H_{56}NO_5$  [M<sup>+</sup>+H]: 594.4158; found 594.4153.

(1R,2S,3S,6S)-1,6-Dihydroxy-2,3-epoxy-1-{(2S)-3-[(4-methoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)-(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}cyclohexane (33 a): A solution of aqueous TBHP (18  $\mu$ L, 0.147 mmol) in toluene (0.2 mL) was dried with molecular sieves  $(4 \text{ Å})$  at room temperature under Ar. After being stirred for 20 min, a solution of 32 a (8.7 mg, 0.0147 mmol) and [VO-  $(\text{acac})_2$ ] (2.1 mg, 0.008 mmol) in dry toluene (0.5 mL) was added to the above mixture at  $0^{\circ}$ C. After being stirred for 30 min, the reaction mixture was quenched with saturated  $Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>$  (aq.) and extracted with Et<sub>2</sub>O. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using hexane/EtOAc  $(3:2)$  as the eluent to give 33a  $(6.1 \text{ mg}, 69\%)$  as a yellow oil.  $[\alpha]_{\text{D}}^{24} = +10.2$  (c=1.10, CHCl<sub>3</sub>); IR (KBr):  $\tilde{\nu} = 3305$ , 2956, 1651, 1606, 1514, 1454 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.79 (3H, d, J=6.6 Hz), 0.83 (3H, t, J=7.1 Hz), 0.90 (3H, d, J=6.8 Hz), 0.99 (3H, d, J=6.6 Hz), 1.12–1.32 (4H, m), 1.52 (3H, s), 1.53–1.67 (4H, m), 1.71– 1.78 (1H, m), 1.84–2.08 (4H, m), 2.23 (1H, m), 2.33 (1H, m), 2.81 (1H, d,  $J=3.5$  Hz), 3.40 (1H, d,  $J=3.5$  Hz), 3.72–3.81 (3H, m), 3.82 (3H, s), 4.21 (1H, m), 4.50 (2H, s), 4.83 (1H, d,  $J=9.3$  Hz), 5.70 (1H, dd,  $J=8.4$ , 14.7 Hz), 5.74 (1H, d, J=14.8 Hz), 6.08 (1H, dd, J=10.8, 14.7 Hz), 6.16

#### **A EUROPEAN JOURNAL**

(1H, dd,  $J=11.4$ , 14.6 Hz), 6.23 (1H, d,  $J=6.5$  Hz), 6.47 (1H, dd,  $J=$ 10.8, 14.6 Hz), 6.90 (2H, d,  $J=8.6$  Hz), 7.22 (1H, dd,  $J=11.4$ , 14.8 Hz), 7.28 ppm (2H, d,  $J=8.6$  Hz); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta = 12.1$ , 15.9, 18.6, 19.9, 21.1, 21.4, 28.3, 30.5, 32.3, 34.1, 34.9, 38.4, 43.9, 46.1, 48.2, 55.3, 56.1, 57.8, 67.0, 69.9, 71.4, 73.0, 113.9, 121.6, 127.6, 128.3, 129.6, 129.9, 132.1, 133.1, 140.8, 142.2, 145.9, 159.6, 166.9 ppm; FAB-MS m/z: 610 [M<sup>+</sup> +H]; HR-FAB-MS: calcd for  $C_{37}H_{56}NO_6$  [M<sup>+</sup>+H]: 610.4108; found 610.4092.

#### (2S,3S,4S)-3,4-Epoxy-2-hydroxy-2-{(2S)-3-[(4-methoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)-(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-

2,4,6,12-tetraenoylamino]propyl}cyclohexan-1-one (34 a): TPAP (1.1 mg, 0,0030 mmol) and NMO  $(4.6 \mu L, 0.0224 \text{ mmol})$  were added to a stirred solution of 33a (9.1 mg, 0.0149 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (0.15 mL) at 0 °C under Ar. After being stirred for 30 min, the reaction mixture was quenched with saturated NaHCO<sub>3</sub> (aq.) and extracted with EtOAc. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated in vacuo. The residue was purified by column chromatography using hexane/EtOAc  $(1:1)$  as the eluent to give 34a  $(4.4 \text{ mg}, 49\%)$  as a yellow oil, and the substrate **33a** (4.0 mg, 44%) was recovered.  $[a]_D^{26} = -7.5$  (c= 1.04, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3290$ , 2923, 1722, 1651, 1610, 1514, 1461 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.77–0.86 (6H, m), 0.90 (3H, d,  $J=6.8$  Hz), 0.99 (3H, d,  $J=6.3$  Hz), 1.10–1.40 (4H, m), 1.52 (3H, s), 1.55 (1H, m), 1.76 (1H, m), 1.87 (1H, m), 2.14 (1H, d, J=2.9Hz), 2.16  $(1\,\text{H}, \text{d}, J = 2.9 \,\text{Hz})$ , 2.21  $(1\,\text{H}, \text{m})$ , 2.30–2.46  $(4\,\text{H}, \text{m})$ , 2.79  $(1\,\text{H}, \text{m})$ , 3.36  $(1\,\text{H}, \text{d}, J=4.1 \text{ Hz})$ , 3.39 (1H, m), 3.45 (1H, dd,  $J=5.3$ , 9.4 Hz), 3.65 (1H, dd,  $J=4.4$ , 9.4 Hz), 3.81 (3H, s), 4.06 (1H, m), 4.13 (1H, br s), 4.38 (1H, d,  $J=11.2$  Hz), 4.43 (1H, d,  $J=11.2$  Hz), 4.84 (1H, d,  $J=9.8$  Hz), 5.69 (1H, dd,  $J=8.5$ , 15.2 Hz), 5.71 (1H, d,  $J=14.9$  Hz), 6.06 (1H, d,  $J=$ 8.5 Hz), 6.08 (1H, dd, J=11.0, 15.2 Hz), 6.16 (1H, dd, J=11.2, 14.9Hz), 6.48 (1 H, dd,  $J=11.0$ , 14.9 Hz), 6.78 (2 H, d,  $J=8.5$  Hz), 7.19 (1 H, dd,  $J=11.2$ , 14.9 Hz), 7.24 ppm (2H, d,  $J=8.5$  Hz); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  = 12.1, 16.2, 19.5, 21.1, 22.3, 24.9, 25.6, 28.3, 29.7, 30.5, 31.6, 33.9, 35.1, 46.3, 49.2, 53.0, 60.7, 70.9, 72.9, 77.8, 113.9, 122.2, 127.7, 128.3, 129.6, 129.9, 132.1, 133.1, 140.5, 141.6, 145.6, 159.4, 166.1, 209.9 ppm; FAB-MS  $m/z$ : 608 [M<sup>+</sup>+H]; HR-FAB-MS: calcd for C<sub>37</sub>H<sub>54</sub>NO<sub>6</sub> [M<sup>+</sup> +H]: 608.3951; found 608.3937.

#### (4S,5S,6S)-4,5-Epoxy-6-hydroxy-6-{(2S)-3-[(4-methoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)-(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-

2,4,6,12-tetraenoylamino]propyl}2-cyclohexen-1-one (35 a): LDA in THF  $(0.25 \text{ m})$  was prepared as follows: *nBuLi*  $(1.56 \text{ m} \text{ in hexane}, 0.78 \text{ mL})$  was added to a solution of diisopropylamine (70.3 µL, 0.5 mmol) in dry THF (1.15 mL) at  $-78^{\circ}$ C for 30 min under Ar. LDA (0.25 m in THF, 0.10 mL) and [15]crown-5 (20  $\mu$ L, 0.10 mmol) were added to a solution of 34a (6.1 mg, 0.010 mmol) in THF (0.1 mL) at  $-78^{\circ}$ C under Ar. After being stirred for 10 min, a solution of N-tert-butylbenzenesulfinimidoyl chloride (22.0 mg, 0.10 mmol) in THF (0.1 mL) was added to the reaction mixture. After being stirred for 30 min, the reaction mixture was quenched with water and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using  $Et<sub>2</sub>O$  as the eluent to give 35 a  $(2.3 \text{ mg}, 0.0037 \text{ mmol}, 38\%)$  as a yellow oil, and the substrate 34a  $(1.6 \text{ mg}, 26\%)$  was recovered.  $[\alpha]_D^{25} = +22.3$   $(c=0.32, \text{CHCl}_3)$ ; IR (KBr):  $\tilde{v} = 3320, 2922, 1693, 1651, 1612, 1514, 1454 \text{ cm}^{-1};$  <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta = 0.71 - 0.91$  (9H, m), 0.99 (3H, d, J = 7.0 Hz), 1.20-1.30 (4H, m), 1.52 (3H, s), 1.55 (1H, m), 1.85–2.05 (4H, m), 2.20 (1H, m), 2.31  $(1\,\text{H}, \text{m})$ , 3.46  $(1\,\text{H}, \text{dd}, J=4.0, 9.0 \text{ Hz})$ , 3.53  $(1\,\text{H}, \text{m})$ , 3.58  $(1\,\text{H}, \text{dd}, J=$ 3.5, 9.0 Hz), 3.69 (1H, d, J=4.0 Hz), 3.81 (3H, s), 4.14 (1H, brs), 4.22 (1H, m), 4.42 (2H, s), 4.84 (1H, d, J=9.0 Hz), 5.69 (1H, m) 5.70 (1H, d,  $J=15.0$  Hz), 5.91 (1H, d,  $J=8.7$  Hz), 6.08 (1H, dd,  $J=11.0$ , 15.5 Hz), 6.16 (1H, dd, J=10.5, 14.0 Hz), 6.17 (1H, d, J=10.5 Hz), 6.48 (1H, dd,  $J=11.0, 14.0$  Hz), 6.88 (2H, d,  $J=8.7$  Hz), 7.08 (1H, dd,  $J=3.5, 8.7$  Hz), 7.22 (1H, m), 7.23 ppm (2H, d,  $J=8.7$  Hz); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): d=11.9, 14.0, 15.9, 21.0, 22.6, 28.2, 29.2, 29.7, 32.1, 33.5, 34.9, 38.3, 44.9, 47.1, 55.3, 58.9, 65.5, 71.4, 72.9, 106.8, 109.1, 113.9, 121.8, 127.2, 128.0, 128.5, 129.5, 130.3, 132.9, 140.0, 141.9, 143.9, 158.9, 165.6, 197.3 ppm; FAB-MS  $m/z$ : 606 [M<sup>+</sup>+H]; HR-FAB-MS: calcd for C<sub>37</sub>H<sub>52</sub>NO<sub>6</sub> [M<sup>+</sup> +H]: 606.3795; found 606.3796.

#### (4S,5S,6S)-4,5-Epoxy-6-hydroxy-6-{(2S)-3-hydroxy-2-[(2E,4E,6E,12E)- (8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}-2-cyclohexen-1-one (scyphostatin (1)):  $Ph_3C^{+}BF_4^-$  (6.6 mg,

0.0200 mmol) was added to a solution of 35 a (10.1 mg, 0.0167 mmol) in  $CH_2Cl_2$  (0.2 mL) at 0 °C under Ar. After being stirred for 5 min, the reaction mixture was quenched with saturated  $NaHCO<sub>3</sub>$  (aq.) and extracted with EtOAc. The organic layer was washed with brine, dried over Na2SO4, and evaporated in vacuo. The residue was purified by column chromatography using EtOAc/acetone (4:1) as the eluent to give scyphostatin (1) (2.67 mg, 32%) as a yellow oil.  $[\alpha]_D^{25} = +60.5$  (c=0.20, MeOH); IR (KBr):  $\tilde{v} = 3292, 2956, 2923, 1697, 1651, 1606, 1512,$ 1454 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD):  $\delta$  = 0.83–0.86 (6H, m), 0.91  $(3H, d, J=6.7 Hz)$ , 1.00  $(3H, d, J=6.8 Hz)$ , 1.04  $(1H, m)$ , 1.19  $(1H, m)$ , 1.27–1.41 (2H, m), 1.54 (3H, s), 1.58 (1H, m), 1.78 (1H, m), 1.86 (1H, m),  $1.88$  (1H, m),  $2.08$  (1H, dd,  $J=3.7$ ,  $14.7$  Hz),  $2.27$  (1H, m),  $2.35$  (1H, m), 3.46 (1H, dd, J=5.5, 11.0 Hz), 3.52 (1H, dd, J=4.9, 11.0 Hz), 3.59 (1H, m), 3.66 (1H, d, J=3.7 Hz), 4.04 (1H, m), 4.86 (1H, m), 5.71 (1H, m), 5.89(1H, d, J=15.3 Hz), 6.07 (1H, dd, J=1.3, 9.8 Hz), 6.15 (1H, dd,  $J=11.6$ , 14.6 Hz), 6.26 (1H, dd,  $J=11.0$ , 15.3 Hz), 6.54 (1H, dd,  $J=11.0$ , 14.6 Hz), 7.09–7.18 ppm (2H, m); <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>OD):  $\delta$  = 12.9, 16.9, 20.5, 22.0, 22.5, 30.1, 32.2, 36.0, 36.7, 40.3, 46.0, 48.5, 49.8, 50.1, 58.8, 65.9, 78.0, 124.2, 129.9, 130.4, 132.6, 134.5, 134.7, 141.8, 142.9, 146.4, 146.8, 169.0, 200.4 ppm; FAB-MS m/z: 486 [M++H]; HR-FAB-MS: calcd for  $C_{29}H_{44}NO_5 [M^+ + H]$ : 486.3219; found 486.3199. The <sup>1</sup>H and <sup>13</sup>C NMR, IR, and HR-FAB-MS spectra showed good agreement with those of authentic sample (see the Acknowledgements).

Experiment in Scheme 8: 2,4,6-Collidine (0.18 mL, 1.38 mmol) and TESOTf (0.21 mL, 0.920 mmol) were successively added to a solution of **24** (46.1 mg, 0.230 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (2.3 mL) at 0 °C under N<sub>2</sub>. After being stirred at the same temperature for 5 min, the reaction mixture was quenched with water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was purified by column chromatography using hexane/EtOAc  $(25:1)$  as the eluent to give 36 (41.1 mg, 0.146 mmol, 63%) as a colorless oil.  $\left[\alpha\right]_D^{22}$  = +136.5 (c = 0.64, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v}$  = 2875, 1445, 1238 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.57 (6H, q, J = 7.9 Hz), 0.93 (9H, t, J = 7.9Hz), 1.80–1.88 (2H, m), 1.95–2.12 (3H, m), 2.33–2.48 (1H, m), 3.35  $(3H, s)$ , 4.04 (1H, t, 5.4 Hz), 5.09 (1H, dd,  $J=3.2$ , 6.3 Hz), 5.70–5.83 ppm  $(2H, m)$ ; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 6.4, 6.9, 21.2, 26.1, 47.8, 55.4,$ 77.2, 83.4, 104.7, 128.2, 130.6 ppm; FAB-MS m/z: 307 [M++Na]; HR-FAB-MS: calcd for  $C_{15}H_{28}O_3$ SiNa [ $M^+$ +Na]: 307.1701; found 307.1736.

General procedure for the one-pot synthesis of disilyl aldehydes: Silyl triflate ( $\rm R^{1}OTH$ ) was added to a solution of 2,4,6-collidine in dry  $\rm CH_{2}Cl_{2}$  at 0 or  $-78$ °C under N<sub>2</sub>. After being stirred at the same temperature for 30 min, a solution of  $24$  in dry  $CH_2Cl_2$  was added to the reaction mixture at the same temperature. After being stirred at the same temperature for 5–10 min, silyl triflate ( $R<sup>2</sup>$ OTf) was added to the reaction mixture at the same temperature. After being stirred at the same temperature for 5 min, the reaction mixture was quenched with water and extracted with  $CH_2Cl_2$ . The organic layer was washed with brine, dried over  $Na_2SO_4$ , and concentrated in vacuo. The residue was purified by column chromatography using hexane/EtOAc (25:1) as the eluent to give 26 in the yield shown in Table 1.

(1S,6S)-1,6-Di(triethylsiloxy)cyclohex-2-en-1-yl acetaldehyde (26 b): Colorless oil;  $\lbrack a \rbrack_{D}^{24} = +38.5$  (c=0.68, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 2955$ , 1715, 1458, 1238 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.61 (12 H, q, J = 7.9 Hz), 0.94 (9H, t, J=7.9Hz), 0.95 (9H, t, J=7.9Hz), 1.57–1.67 (1H, m), 1.84– 1.94 (1H, m), 1.99–2.19 (2H, m), 2.33 (1H, dd, J=2.4, 15.9Hz), 2.59  $(1\,\text{H}, \text{dd}, J = 3.0, 15.9 \text{ Hz})$ , 3.90  $(1\,\text{H}, \text{dd}, J = 3.2, 9.2 \text{ Hz})$ , 5.59  $(1\,\text{H}, \text{dt}, J =$ 10.1, 1.8 Hz), 5.78 (1 H, dt,  $J=10.1$ , 3.6 Hz), 9.79 ppm (1 H, t,  $J=3.0$  Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.9, 6.7, 6.8, 7.1, 23.1, 28.3, 51.7, 75.5, 75.7, 129.6, 131.0, 202.3 ppm; FAB-MS m/z: 407 [M++Na]; HR-FAB-MS: calcd for  $C_{20}H_{40}O_3Si_2Na$  [ $M^+ +Na$ ]: 407.2414; found 407.2433.

(1S,6S)-1,6-Di(tert-butyldimethylsiloxy)cyclohex-2-en-1-yl acetaldehyde (26c): Colorless oil;  $\left[\alpha\right]_D^{25} = +64.4$  (c=0.70, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 2955$ , 1717, 1471, 1256 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.07 (3H, s), 0.08 (3H, s), 0.09(3H, s), 0.11 (3H, s), 0.86 (18H, s), 1.58–1.69(1H, m), 1.89– 2.06 (2H, m), 2.12–2.21 (1H, m), 2.35 (1H, dd, J=2.7, 15.9Hz), 2.67

 $(1\,\text{H}, \text{dd}, J = 2.7, 15.9 \text{ Hz})$ , 3.92  $(1\,\text{H}, \text{dd}, J = 2.4, 7.5 \text{ Hz})$ , 5.56  $(1\,\text{H}, \text{d}, J =$ 10.1 Hz), 5.83 (1H, dt,  $J=10.1$ , 3.6 Hz), 9.83 ppm (1H, t,  $J=2.7$  Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = -4.7, -4.2, -2.3, -1.9, 18.0, 18.2, 22.2,$ 25.8, 25.9, 27.1, 51.7, 74.5, 74.8, 130.2, 130.4, 202.7 ppm; FAB-MS m/z: 407  $[M^+ + Na]$ ; HR-FAB-MS: calcd for  $C_{20}H_{40}O_3Si_2Na$   $[M^+ + Na]$ : 407.2414; found 407.2415.

(1S,6S)-1-Trimethylsiloxy-6-(tert-butyldimethylsiloxy)cyclohex-2-en-1-yl **acetaldehyde (26d):** Colorless oil;  $[\alpha]_D^{25} = +41.0$  ( $c = 1.20$ , CHCl<sub>3</sub>): IR (KBr):  $\tilde{v} = 2929, 1722, 1461, 1251 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta =$ 0.07 (6H, s), 0.12 (9H, s), 0.86 (9H, s), 1.59 (1H, m), 1.84 (1H, m) 1.96– 2.17 (2H, m), 2.30 (1H, dd,  $J=2.6$ , 15.7 Hz), 2.65 (1H, dd,  $J=3.0$ , 15.7 Hz), 3.86 (1H, dd,  $J=3.3$ , 9.5 Hz), 5.61 (1H, dt,  $J=10.0$ , 1.8 Hz), 5.77 (1H, dt,  $J=10.0$ , 3.5 Hz), 9.81 ppm (1H, t,  $J=3.0$  Hz); <sup>13</sup>C NMR  $(75 \text{ MHz}, \text{ CDCl}_3): \delta = -4.6, -4.5, 2.4, 18.0, 23.3, 25.8, 28.2, 51.2, 75.4,$ 76.3, 129.6, 130.7, 202.9 ppm; FAB-MS  $m/z$ : 365 [M<sup>+</sup>+Na], HR-FAB-MS  $m/z$ : calcd for C<sub>17</sub>H<sub>34</sub>O<sub>3</sub>Si<sub>2</sub>Na [M<sup>+</sup>+Na]: 365.1944; found 365.1949.

(1S,6S)-1-Triisopropylsiloxy-6-triethylsiloxy-cyclohex-2-en-1-yl acetaldehyde (26e): Colorless oil;  $[\alpha]_D^{21} = +15.6$  (c=0.52, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} =$ 2947, 1715, 1462, 1101 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.62 (6 H, q,  $J=7.8$  Hz), 0.94 (9H, t,  $J=7.8$  Hz), 1.08 (18H, s), 0.88–1.15 (3H, m), 1.58–1.69(1H, m), 1.87–1.96 (1H, m), 2.02–2.18 (2H, m), 2.34 (1H, dd,  $J=2.4$ , 15.8 Hz), 2.69 (1H, dd,  $J=3.4$ , 15.8 Hz), 4.06 (1H, dd,  $J=3.3$ , 10.5 Hz), 5.60 (1H, dt,  $J=9.8$ , 1.7 Hz), 5.74 (1H, dt,  $J=9.8$ , 3.7 Hz), 9.91 ppm (1H, dd,  $J=2.4$ , 3.4 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 6.8$ , 7.1, 12.9, 18.15, 18.23, 24.1, 29.4, 51.2, 77.1, 77.2, 128.9, 131.6, 203.1 ppm; FAB-MS  $m/z$ : 449 [M<sup>+</sup>+Na]; HR-FAB-MS: calcd for C<sub>23</sub>H<sub>46</sub>O<sub>3</sub>Si<sub>2</sub>Na  $[M^+ + Na]$ : 449.2844; found 449.2880.

Experiments in Table 2: Compounds 38 a–c were prepared from cyclohexylacetoaldehyde and the corresponding stannyl compound with 4-methoxyphenylmethyloxymethyl (for 38 a), 3,4-dimethoxyphenylmethyloxymethyl (for  $38b$ ), or 2.4-dimethoxyphenylmethyloxymethyl (for  $38c$ ) groups in the same way as compounds 31 a and 31 b (see Schemes 7 and 11, respectively).

MPM ether (38a): White crystals; m.p. 110–112°C; IR (KBr):  $\tilde{v} = 3273$ , 2923, 1651, 1612, 1514, 1448 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.88  $(3H, t, J=6.2 \text{ Hz})$ , 0.85–0.98 (2H, m), 1.08–1.32 (15H, m), 1.35–1.49  $(2H, m), 1.52-1.71$   $(7H, m), 1.74-1.83$   $(1H, m), 2.12$   $(2H, dd, J=6.8,$ 14.3 Hz), 3.47 (2H, t, J=3.1 Hz), 3.81 (3H, s), 4.27 (1H, m), 4.40 (1H, d, J=11.8 Hz), 4.47 (1H, d, J=11.8 Hz), 5.56 (1H, d, J=9.3 Hz), 5.75 (1H, d,  $J=14.9$  Hz), 5.88 (1H, m), 6.07–6.22 (2H, m), 6.49 (1H, dd,  $J=10.6$ , 14.9Hz), 6.68 (2H, d, J=8.7 Hz), 7.22 (1H, m), 7.24 ppm (2H, d, J= 8.7 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 14.1, 22.6, 26.1, 26.2, 26.5, 29.0, 29.2, 29.3, 29.4, 29.5, 31.8, 33.0, 33.6, 34.3, 39.5, 46.5, 55.3, 71.8, 72.9, 76.5, 113.8, 122.8, 127.8, 129.3, 129.8, 130.2, 139.6, 139.9, 141.0, 159.2, 165.5 ppm; FAB-MS  $m/z$ : 510  $[M^+ + H]$ ; HR-FAB-MS: calcd for  $C_{33}H_{52}NO_3$  [ $M^+$ +H]: 510.3947; found 510.3946.

<sup>3,4</sup>DMPM ether (38b): White crystals; m.p. 74–75 °C; IR (KBr):  $\tilde{v} = 3284$ , 2923, 1651, 1608, 1514, 1448 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.88 (3H, t, J=7.0 Hz), 0.88–0.99 (2H, m), 1.11–1.32 (15H, m), 1.35–1.46  $(3H, m)$ , 1.57–1.70 (6H, m), 1.74–1.83 (1H, m), 2.11 (2H, dd,  $J=6.9$ , 14.1 Hz), 3.46 (2H, d,  $J=4.0$  Hz), 3.80 (6H, s), 4.29 (1H, m), 4.39 (1H, d,  $J=11.5$  Hz), 4.47 (1H, d,  $J=11.5$  Hz), 5.57 (1H, d,  $J=8.8$  Hz), 5.76 (1H, d,  $J=14.7$  Hz), 5.83-5.92 (1H, m), 6.06-6.21 (2H, m), 6.48 (1H, dd,  $J=$ 10.6, 14.5 Hz), 6.80–6.85 (3H, m), 7.22 ppm (1H, dd, J=11.2, 14.7 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 14.0, 22.6, 26.1, 26.4, 28.9, 29.1, 29.2, 29.3, 29.4, 31.8, 32.9, 33.5, 34.2, 39.5, 46.5, 55.7, 55.8, 71.8, 73.0, 73.1, 110.7, 110.8, 110.9, 120.2, 122.7, 127.6, 129.7, 130.6, 139.5, 139.9, 141.0, 148.5, 148.9, 165.5 ppm; FAB-MS m/z: 540 [M++H]; HR-FAB-MS: calcd for  $C_{34}H_{54}NO_4$  [ $M^+$ +H]: 540.4053; found 540.4039.

<sup>2,4</sup>DMPM ether (38 c): White crystals; m.p. 67–68 °C; IR (KBr):  $\tilde{v} = 3284$ , 2923, 1651, 1612, 1510, 1454 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.88 (3H, t, J=7.1 Hz), 0.85–0.99 (2H, m), 1.11–1.30 (15H, m), 1.38–1.45  $(3H, m)$ , 1.57–1.71 (6H, m), 1.74–1.84 (1H, m), 2.12 (2H, dd,  $J=6.9$ , 14.1 Hz), 3.49(2H, d, J=3.3 Hz), 3.79(6H, s), 4.27 (1H, m), 4.40 (1H, d,  $J=11.7$  Hz), 4.50 (1H, d,  $J=11.7$  Hz), 5.78 (1H, d,  $J=14.8$  Hz), 5.88 (2H, m), 6.06–6.21 (2H, m), 6.42–6.50 (3H, m), 7.17–7.22 ppm (2H, m); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = 13.9, 22.4, 26.0, 26.3, 28.8, 28.9, 29.0, 29.2, 29.3, 31.6, 32.7, 33.3, 34.0, 39.3, 46.3, 55.0, 67.7, 71.5, 98.2, 98.3, 103.6, 103.7, 118.6, 122.9, 127.7, 129.6, 130.2, 130.3, 139.1, 139.4, 140.5, 158.3, 160.4, 165.3 ppm; FAB-MS m/z: 540 [M++H]; HR-FAB-MS: calcd for  $C_{34}H_{54}NO_4$  [ $M^+$ +H]: 540.4053; found 540.4058.

General procedure for the deprotection of MPM-type ethers with  $Ph_3C^+$  $BF_4^-$ :  $Ph_3C^+BF_4^-$  (1.1 mmol) was added to a solution of MPM-type ether 38 (1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) at 0<sup>o</sup>C under Ar. After completion of the reaction (TLC check), the reaction mixture was quenched by the addition of saturated NaHCO<sub>3</sub> (aq.), and the reaction mixture was extracted with  $CH_2Cl_2$ . The organic layer was dried over  $Na_2SO_4$  and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/EtOAc (10:1) as the eluent to give 39 in the yield shown in Table 2.

**Alcohol (39):** White crystals; m.p. 84–86 °C; IR (KBr):  $\tilde{v} = 3280, 2923,$ 1651, 1608, 1539, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.88 (3 H, t, J=7.0 Hz), 0.87–1.00 (2H, m), 1.11–1.32 (17H, m), 1.34–1.44 (2H, m), 1.58–1.83 (6H, m), 2.12 (2H, dd,  $J=6.8$ , 14.3 Hz), 3.01 (1H, brs), 3.56  $(1\,\text{H}, \text{dd}, J = 6.2, 10.9 \text{ Hz}),$  3.72  $(1\,\text{H}, \text{dd}, J = 3.1, 10.9 \text{ Hz}),$  4.15  $(1\,\text{H}, \text{m}),$ 5.51 (1H, d, J=7.5 Hz), 5.81–5.95 (2H, m), 6.07–6.22 (2H, m), 6.51 (1H, dd, J=10.6, 14.3 Hz), 7.25 ppm (1H, dd, J=10.5, 14.9Hz); 13C NMR  $(75 \text{ MHz}, \text{ CDCl}_3): \delta = 14.1, 22.6, 26.0, 26.2, 26.4, 28.9, 29.2, 29.3, 29.5,$ 31.8, 32.8, 32.9, 33.7, 34.3, 38.8, 49.8, 67.0, 76.6, 121.9, 127.5, 129.7, 140.2, 140.6, 141.9, 167.2 ppm; FAB-MS  $m/z$ : 390 [M<sup>+</sup>+H]; HR-FAB-MS: calcd for  $C_{25}H_{44}NO_2$  [ $M^+$ +H]: 390.3372; found 390.3390.

Preparation of <sup>2,4</sup>DMPMOCH<sub>2</sub>SnBu<sub>3</sub> (40):<sup>[11]</sup> A mixture of 2,4-methoxybenzyl alcohol (2.54 g, 15.1 mmol) and NaH (0.70 g, 60% in oil, 17.4 mmol) in THF (40 mL) was stirred at room temperature under  $N_2$ . After being stirred for 1 h at the same temperature, a solution of tri-n-butylstannylmethyl iodide (5.0 g, 11.6 mmol) in THF (8 mL) was added slowly to the solution. The resulting mixture was stirred for three days. The reaction mixture was quenched by addition of MeOH, and the solution was diluted with  $Et_2O$ . The organic layer was washed with water two times, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The residue was purified by column chromatography on silica gel using hexane/Et<sub>2</sub>O (30:1) as the eluent to give 40 (3.7 g, 68%) as a colorless oil. IR (KBr):  $\tilde{v} =$ 2954, 1614, 1506, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.85–0.94 (15H, m), 1.28 (6H, m), 1.51 (6H, m), 3.72 (2H, s), 3.87 (3H, s), 3.88 (3H, s), 4.35 (2H, s), 6.83–6.87 ppm (3H, m); 13C NMR (75 MHz, CDCl3) d: 9.0, 13.7, 27.3, 29.1, 55.3, 61.4, 71.5, 98.3, 103.7, 119.7, 129.7, 158.2, 160.2 ppm.

(2R)-1-[(1S,6S)-1-Trimethylsiloxy-6-(tert-butyldimethylsiloxy)cyclohex-2 en-1-yl]-3-[(2,4-dimethoxyphenyl)methyloxy]propan-2-ol (27b) and (2S)- 1-[(1S,6S)-1-Trimethylsiloxy-6-(tert-butyldimethylsiloxy)cyclohex-2-en-1-

yl]-3-[(2,4-dimethoxyphenyl)methyloxy]propan-2-ol (27 b'): nBuLi (1.58m in hexane, 0.24 mL) was added dropwise to a solution of <sup>2,4</sup>DMPMOCH<sub>2</sub>SnBu<sub>3</sub> (40; 177.2 mg, 0.376 mmol) in dry THF (0.9 mL) at  $-78$ °C under N<sub>2</sub>. After being stirred for 15 min, a solution of 26d  $(36.1 \text{ mg}, 0.105 \text{ mmol})$  in THF  $(0.9 \text{ mL})$  was added to the reaction mixture. After being stirred for 5 min, the reaction mixture was quenched with water and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated in vacuo. The residue was purified by column chromatography using hexane/EtOAc (10:1) as the eluent to give 27b  $(30.8 \text{ mg}, 0.0588 \text{ mmol}, 56\%)$  and 27b'  $(12 \text{ mg},$ 0.0378 mmol, 25%).

**27b**: Colorless oil;  $[a]_D^{25} = +26.0$  ( $c = 0.67$ , CHCl<sub>3</sub>): IR (KBr):  $\tilde{v} = 3498$ , 2945, 1614, 1504, 1454 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.11 (3 H, s), 0.12 (3H, s), 0.16 (9H, s), 0.88 (9H, s), 1.62 (1H, dd, J=9.5, 14.3 Hz), 0.74 (1H, m), 1.84–1.97 (2H, m), 2.04 (1H, m), 2.14 (1H, m), 3.45 (2H, d,  $J=5.3$  Hz), 3.83 (6H, s), 3.89 (1H, br s), 3.95 (1H, dd,  $J=2.2$ , 7.5 Hz), 4.22 (1H, m), 4.57 (2H, s), 5.68 (1H, d,  $J=10.1$  Hz), 5.81 (1H, dt,  $J=$ 10.1, 3.1 Hz), 6.47 (1H, s), 6.49 (1H, d, J=7.9 Hz), 7.29 ppm (1H, d, J= 7.9 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = -4.8, -4.1, 2.3, 17.9, 22.5, 25.7, 26.6, 40.6, 55.2, 67.2, 67.8, 73.5, 74.9, 98.3, 103.8, 119.3, 129.6, 130.2, 131.1, 158.4, 160.4 ppm; FAB-MS m/z: 547 [M++Na]; HR-FAB-MS: calcd for  $C_{27}H_{48}O_6Si_2Na$  [ $M^+$ +Na]: 547.2887; found 547.2890.

**27b'**: Colorless oil;  $[a]_D^{26} = -8.4$  (c=0.64, CHCl<sub>3</sub>): IR (KBr):  $\tilde{\nu} = 3512$ , 2931, 1614, 1508, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 0.07$  (3H, s), 0.08 (3H, s), 0.11 (9H, s), 0.90 (9H, s), 1.63 (1H, dd,  $J=2.4$ , 14.9 Hz), 1.63–1.83 (2H, m), 2.01 (1H, dd, J=9.2, 14.9 Hz), 2.05–2.12 (2H, m),

3.42 (1 H, dd,  $J=5.5$ , 9.7 Hz), 3.47 (1 H, dd,  $J=6.0$ , 9.7 Hz), 3.78 (3 H, s), 3.79 (3H, s), 3.83 (1H, dd,  $J=3.7$ , 10.4 Hz), 4.02 (1H, br s), 4.13 (1H, m), 4.52 (2H, s), 5.65 (2H, dt, J=10.1, 2.9Hz), 6.43 (1H, s), 6.45 (1H, m), 7.25 ppm (1H, d,  $J=8.3$  Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = -4.6, -4.4,$ 2.5, 24.2, 25.8, 28.6, 41.1, 55.3, 66.4, 67.8, 74.9, 76.6, 78.1, 98.3, 103.8, 119.3, 127.9, 130.2, 132.1, 158.4, 160.4 ppm; FAB-MS m/z: 547 [M++Na]; HR-FAB-MS: calcd for  $C_{27}H_{48}O_6Si_2Na$  [ $M^+$ +Na]: 547.2887; found 547.2889.

Determination of the absolute configurations of the secondary alcohols of 27b and 27b': The absolute configurations of the secondary alcohols of  $27b$  and  $27b'$  were determined by the modified Mosher's method.<sup>[12]</sup> Each MTPA ester was synthesized in the same way as 27 a and 27 b.



General procedure for the preparation of an MTPA ester: DCC (3.3 mmol), MTPA (4.3 mmol), and DMAP (3.3 mmol) were added successively to a stirred solution of secondary alcohol 27b or 27b'  $(1.0 \text{ mmol})$  in CH<sub>2</sub>Cl<sub>2</sub>  $(10.0 \text{ mL})$  at room temperature. After being stirred for 30 min, the solvent was removed in vacuo. The residue was purified by column chromatography using hexane/EtOAc (20:1) as the eluent.

(2R)-1-[(1S,6S)-1-Trimethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2 en-1-yl]-3-[(2,4-dimethoxyphenyl)methyloxy]-propan-2-yl (2S)-3,3,3-trifluoro-2-methoxy-2-phenylpropanoate (27b; (S)-MTPA): Colorless oil;



 $[\alpha]_D^{26} = -27.5$  (c=1.03, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} =$ 2935, 1745, 1614, 1508, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{ CDCl}_3): \delta = 0.07 \text{ (6H, s)}, 0.13$ (9H, s), 0.87 (9H, s), 1.57 (1H, m), 1.72 (1H, m), 1.74 (1H, dd, J=4.7, 15.2 Hz), 2.07 (2H, m), 2.14 (1H, dd, J=5.7, 15.2 Hz), 3.49(3H, s), 3.51 (1H, m), 3.63 (1H, dd, J=2.4, 10.6 Hz), 3.73 (1H, m), 3.76 (3H, s), 3.80

 $(3H, s)$ , 4.40  $(1H, d, J=12.1 Hz)$ , 4.46  $(1H, d, J=12.1 Hz)$ , 5.69  $(1H, m)$ , 5.72 (1H, m), 5.79 (1H, d,  $J=10.1$  Hz), 6.42 (2H, m), 7.17 (1H, d,  $J=$ 8.8 Hz), 7.24–7.43 (3H, m), 7.58 ppm (2H, d, J=7.5 Hz).

(2R)-1-[(1S,6S)-1-Trimethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2 en-1-yl]-3-[(2,4-dimethoxyphenyl)methyloxy]-propan-2-yl (2R)-3,3,3-trifluoro-2-methoxy-2-phenylpropanoate (27b; (R)-MTPA): Colorless oil;



 $[\alpha]_D^{24} = -8.4$  (c=0.91, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} =$ 2935, 1745, 1614, 1508, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 0.04$  (6H, s), 0.13 (9H, s), 0.84 (9H, s), 1.50 (1H, m), 1.59 (1H, dd,  $J=4.9$ , 15.2 Hz), 1.71 (1H, dd,  $J=4.0$ , 13.2 Hz), 1.97–2.06 (2H, m), 2.07 (1H, dd,  $J=5.5, 15.2$  Hz), 3.56 (1H, m), 3.57 (3H, m), 3.70 (1H, dt,  $J=10.8$ , 3.7 Hz), 3.76 (1H, m),

3.77 (3H, s), 3.80 (3H, s), 4.48 (1H, d,  $J=11.7$  Hz), 4.55 (1H, d,  $J=$ 11.7 Hz), 5.48 (1 H, dt,  $J=10.0$ , 2.9 Hz), 5.69 (1 H, d,  $J=10.0$  Hz), 5.70 (1H, m), 6.43 (1H, s), 6.44 (1H, d, J=7.0 Hz), 7.21–7.37 (4H, m), 7.59 ppm (2H, d,  $J=7.5$  Hz).

(2S)-1-[(1S,6S)-1-Trimethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2 en-1-yl]-3-[(2,4-dimethoxyphenyl)methyloxy]-propan-2-yl (2S)-3,3,3-tri-



fluoro-2-methoxy-2-phenylpropanoate (27 b'; (S)-MTPA): Colorless oil;  $[a]_D^{25} = +1.8$  (c= 1.30, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 2933, 1747, 1614,$ 1508, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$ =0.07 (6H, s), 0.12 (9H, s), 0.90 (9H, s), 1.49 (1H, m), 1.70 (1H, m), 1.74 (1H, dd,  $J=$ 

7.0, 14.8 Hz), 1.89–2.06 (2H, m), 2.10 (1H, dd, J=5.0, 14.8 Hz), 3.56 (3H, s), 3.55–3.64 (2H, m), 3.75 (3H, s), 3.77 (1H, dd, J=2.7, 8.4 Hz), 3.80  $(3H, s)$ , 5.46 (1H, d,  $J=11.9$  Hz), 5.49 (1H, d,  $J=10.0$  Hz), 5.54 (1H, d, J=11.9Hz), 5.65 (1H, dt, J=10.0, 3.5 Hz), 5.73 (1H, m), 6.42 (1H, s), 6.44 (1H, m), 7.22–7.35 (4H, m), 7.59 ppm (2H, d,  $J=7.3$  Hz).

(2S)-1-[(1S,6S)-1-Trimethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2 en-1-yl]-3-[(2,4-dimethoxyphenyl)methyloxy]-propan-2-yl (2R)-3,3,3-trifluoro-2-methoxy-2-phenylpropanoate (27b'; (R)-MTPA): Colorless oil;

 $[\alpha]_D^{25}$  = +35.2 (c = 1.10, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v}$  = 2935, 1745, 1614, 1508, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 0.07$  (6H, s), 0.12 (9H, s), 0.90 (9H, s), 1.64 (1H, m), 1.78 (1H, m), 1.80 (1H, dd, J=7.0, 14.6 Hz), 1.95–2.15  $(2H, m)$ , 2.21  $(1H, dd, J=4.2, 14.6 Hz)$ , 3.52 (1H, dd, J=8.0, 11.2 Hz), 3.54 (3H, s), 3.67  $(1\,\text{H}, \text{dd}, J = 3.0, 11.2 \text{ Hz})$ , 3.76  $(3\,\text{H}, \text{s})$ , 3.80



 $(3H, s)$ , 3.83 (1H, dd,  $J=3.3$ , 9.7 Hz), 4.39 (1H, d,  $J=12.0$  Hz), 4.45 (1H, d,  $J=12.0$  Hz), 5.63 (1H, d,  $J=10.2$  Hz), 5.71 (1H, dt,  $J=10.2$ , 3.3 Hz), 5.76 (1H, m), 6.41 (1H, s), 6.43 (1H, m), 7.17–7.47 (4H, m), 7.57 ppm  $(2H, d, J=7.3 Hz).$ 

(2S)-2-Azido-1-[(1S,6S)-1-trimethylsiloxy-6-tert-butyldimethylsiloxycyclohex-2-en-1-yl]-3- $[(2,4-dimethoxyphenyl)$ methyloxy]propane  $(28 b)$ : PPh<sub>3</sub> (122.0 mg, 0.465 mmol), DEAD (0.21 mL, 40% in toluene, 0.465 mmol), and DPPA (50  $\mu$ L, 0.223 mmol) were added successively to a stirred solution of 27 b (81.4 mg, 0.155 mmol) in THF (1.6 mL) at room temperature under Ar. The reaction mixture was stirred at the same temperature for 30 min. After removal of the solvent in vacuo, the residue was purified by column chromatography using hexane/EtOAc (17:1) as the eluent to give **28b** (64.0 mg, 0.116 mmol, 75%) as a colorless oil.  $[a]_D^{25} = +18.8$  ( $c =$ 0.77, CHCl<sub>3</sub>): IR (KBr):  $\tilde{v} = 2933$ , 2108, 1614, 1508, 1463 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(300 \text{ MHz}, \text{CDCl}_3): \delta = 0.08 \text{ (6H, s)}, 0.10 \text{ (9H, s)}, 0.88 \text{ (9H, s)}, 1.56 \text{ (1H, s)}$ dd,  $J=5.5$ , 14.6 Hz), 1.63 (1H, m), 1.83 (1H, m), 2.01 (1H, dd,  $J=6.8$ , 14.6 Hz), 2.02 (1H, m), 2.12 (1H, m), 3.48 (1H, dd, J=7.9, 9.9 Hz), 3.66  $(1\,\text{H}, \text{dd}, J = 3.7, 9.9 \text{ Hz})$ , 3.75  $(1\,\text{H}, \text{m})$ , 3.79  $(3\,\text{H}, s)$ , 3.80  $(3\,\text{H}, s)$ , 3.83 (1H, dd, J=2.8, 8.4 Hz), 4.53 (2H, s), 5.57 (1H, d, J=10.1 Hz), 5.76 (1H, dt,  $J=10.1$ , 3.7 Hz), 6.43–6.50 (2H, m), 7.28 ppm (1H, d,  $J=8.0$  Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  = -4.7, -4.2, 2.4, 2.6, 18.1, 22.8, 25.9, 27.3, 38.4, 55.4, 57.7, 67.7, 73.9, 75.1, 98.3, 103.9, 118.9, 129.8, 129.9, 130.8, 158.3, 160.5 ppm; FAB-MS m/z: 572 [M++Na]; HR-FAB-MS: calcd for  $C_{27}H_{47}N_3O_5Si_2Na$  [ $M^+ + Na$ ]: 572.2952; found 572.2939.

(1S,6S)-1-{(2S)-3-[(2,4-Dimethoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)- (8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylami-

no]propyl}-2-cyclohexene-1,6-diol  $(32 b)$ : A solution of  $28 b$   $(275.8 mg,$ 0.50 mmol) in dry THF (5.0 mL) was added to a solution of  $LiAlH<sub>4</sub>$ (190.0 mg, 5.0 mmol) in THF (5.0 mL) at  $0^{\circ}$ C under Ar. After being stirred for 1.5 h at room temperature, the reaction mixture was quenched with water and 15% NaOH (aq.), then the precipitate was filtered through a Celite pad. The filtrate was evaporated in vacuo to give crude 29 b. Side-chain carboxylic acid 30 (168.0 mg, 0.58 mmol), DCC  $(412.7 \text{ mg}, 2.0 \text{ mmol})$ , and DMAP  $(244.3 \text{ mg}, 2.0 \text{ mmol})$  were added to a solution of crude 29b in  $CH_2Cl_2$  (2.5 mL) at room temperature under Ar. After being stirred for 1 h, the reaction mixture was quenched with water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo to give crude coupled product. TBAF (5.0 mL, 1.0m in THF, 5.0 mmol) was added to the crude coupled product in THF (5.0 mL) at room temperature under Ar. After being stirred for 10 h, the reaction mixture was quenched with water and extracted with EtOAc. The organic layer was washed with brine, dried over Na2SO4, and evaporated in vacuo. The residue was purified by column chromatography using hexane/EtOAc (1:2) as the eluent to give **32b** (183.3 mg, 0.295 mmol, 66%) as a yellow oil.  $[\alpha]_D^{24} = +9.81$  ( $c = 2.38$ , CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3305$ , 2923, 1591, 1504, 1454 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(500 \text{ MHz}, \text{CDC1}_3)$ :  $\delta = 0.80$  (3H, d, J = 6.1 Hz), 0.84 (3H, t, J = 7.4 Hz), 0.90 (3H, d,  $J=6.7$  Hz), 0.99 (3H, d,  $J=6.1$  Hz), 1.02 (1H, m), 1.20 (1H, m), 1.27–1.38 (2H, m), 1.52 (3H, s), 1.56 (1H, m), 1.70–1.76 (2H, m), 1.81 (1H, dd,  $J=6.7$ , 14.7 Hz), 1.85-1.93 (2H, m), 2.01 (1H, dd,  $J=3.7$ , 15.3 Hz), 2.07 (1H, m), 2.17 (1H, m), 2.23 (1H, m), 2.32 (1H, m), 3.55  $(1H, brs), 3.62$   $(2H, d, J=4.3 Hz), 3.81$   $(3H, s), 3.82$   $(3H, s), 3.92$   $(1H,$ 

d,  $J=6.7$  Hz), 4.10 (1H, brs), 4.25 (1H, m), 4.47 (1H, d,  $J=11.3$  Hz), 4.52 (1H, d,  $J=11.3$  Hz), 4.83 (1H, d,  $J=9.2$  Hz), 5.51 (1H, d,  $J=$ 9.8 Hz), 5.65–5.72 (2H, m), 5.75 (1H, d,  $J=14.0$  Hz), 6.07 (1H, dd,  $J=$ 11.0, 15.3 Hz), 6.16 (1H, dd,  $J=11.0$ , 14.6 Hz), 6.39 (1H, d,  $J=6.1$  Hz), 6.43–6.50 (3H, m), 7.18–7.24 ppm (2H, m); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  = 12.1, 16.2, 19.5, 21.1, 21.4, 22.9, 25.6, 28.3, 30.5, 34.1, 34.9, 39.2, 44.0, 45.9, 48.3, 49.1, 55.5, 66.1, 72.1, 72.2, 72.4, 98.7, 104.0, 118.5, 122.1, 127.8, 128.2, 129.0, 130.9, 131.5, 132.2, 133.1, 140.5, 141.7, 145.6, 158.8, 161.0, 166.8 ppm; FAB-MS  $m/z$ : 624  $[M^+ + H]$ ; HR-FAB-MS: calcd for  $C_{38}H_{58}NO_6$  [ $M^+$ +H]: 624.4264; found 624.4250.

(1R,2S,3S,6S)-1,6-Dihydroxy-2,3-epoxy-1-{(2S)-3-[(2,4-dimethoxyphenyl) methyloxy]-2-[(2E,4E,6E,12E)-(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}cyclohexane (33 b): A solution of aqueous TBHP (0.46 mL, 3.69mmol) in toluene (3.0 mL) was dried with molecular sieves  $(4 \text{ Å})$  at room temperature under Ar. After being stirred for 20 min, a solution of  $32b$  (230.0 mg, 0.369 mmol) and [VO- $(\text{acac})$ ] (49.1 mg, 0.185 mmol) in dry toluene (3.0 mL) was added to the above mixture at  $0^{\circ}$ C. After being stirred for 30 min, the reaction mixture was quenched with saturated  $Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>$  (aq.) and extracted with Et<sub>2</sub>O. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using hexane/EtOAc  $(3:2)$  as the eluent to give 33b  $(172.9 \text{ mg})$ , 0.269 mmol, 69%) as a yellow oil.  $\left[\alpha\right]_D^{24} = +19.1$  ( $c = 0.73$ , CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3330, 2923, 1651, 1591, 1504, 1454 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta = 0.80$  (3H, d, J = 6.7 Hz), 0.83 (3H, t, J = 7.4 Hz), 0.90 (3H, d,  $J=6.7$  Hz), 0.99 (3H, d,  $J=8.6$  Hz), 1.02 (1H, m), 1.20 (1H, m), 1.28– 1.35 (2H, m), 1.52 (3H, s), 1.52–1.62 (2H, m), 1.74 (1H, m), 1.84–1.97 (4H, m), 2.05 (1H, d, J=14.7 Hz), 2.17 (1H, m), 2.23 (1H, m), 2.32 (1H, m), 2.64 (1H, brs), 2.80 (1H, d, J = 3.1 Hz), 3.39 (1H, m), 3.75–3.80 (3H, m), 3.81 (3H, s), 3.83 (3H, s), 4.21 (1H, m), 4.51 (1H, d, J=11.3 Hz), 4.54 (1H, d,  $J=11.3$  Hz), 4.84 (1H, d,  $J=9.2$  Hz), 5.70 (1H, dd,  $J=8.5$ , 15.3 Hz), 5.74 (1H, d,  $J=14.7$  Hz), 6.08 (1H, dd,  $J=11.0$ , 15.3 Hz), 6.16  $(1\,\text{H}, \text{dd}, J=11.6, 14.7 \text{ Hz}), 6.43 \ (1\,\text{H}, \text{d}, J=6.7 \text{ Hz}), 6.44-6.51 \ (3\,\text{H}, \text{m}),$ 7.19–7.24 ppm (2H, m); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  = 12.1, 16.2, 18.7, 19.5, 19.9, 21.0, 21.4, 28.3, 30.5, 34.1, 35.0, 38.3, 44.0, 46.1, 48.3, 49.1, 55.4, 56.0, 57.9, 67.6, 68.3, 69.9, 71.4, 98.7, 103.9, 118.6, 121.8, 127.7, 128.2, 131.0, 133.1, 135.5, 140.7, 141.9, 145.7, 158.8, 161.0, 166.9 ppm; FAB-MS  $m/z$ : 640 [M<sup>+</sup>+H]; HR-FAB-MS: calcd for  $C_{38}H_{58}NO_7$  [M<sup>+</sup>+ H]: 640.4213; found 640.4220.

(2S,3S,4S)-3,4-Epoxy-2-hydroxy-2-{(2S)-3-[(2,4-dimethoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)-(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}cyclohexan-1-one (34b): Dess-Martin periodinane (120.5 mg, 0.284 mmol) was added to a stirred solution of 33b (90.8 mg, 0.142 mmol) in  $CH_2Cl_2$  (7.0 mL) under N<sub>2</sub>. After being stirred at  $40^{\circ}$ C for 10 min, the reaction mixture was quenched with saturated NaHCO<sub>3</sub> (aq.) and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using hexane/EtOAc (1:1) as the eluent to give  $34b$  (62.5 mg, 0.098 mmol, 69%) as a yellow oil, and the substrate **33b** (10.9 mg, 12%) was recovered.  $[\alpha]_D^{25} = -6.2$  (c=2.52, CHCl<sub>3</sub>); IR (KBr):  $\tilde{v} = 3290, 2923, 1720, 1651, 1612, 1508, 1454 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.80 (3 H, d, J = 6.7 Hz), 0.83 (3 H, t, J = 7.4 Hz), 0.90 (3H, d,  $J=6.7$  Hz), 0.99 (3H, d,  $J=6.7$  Hz), 1.02 (1H, m), 1.20 (1H, m), 1.25–1.36 (2H, m), 1.52 (3H, s), 1.55 (1H, m), 1.75 (1H, m), 1.88 (1H, m), 2.10 (1H, dd,  $J=6.1$ , 15.0 Hz), 2.17 (1H, dd,  $J=6.7$ , 15.0 Hz), 2.24 (1H, m), 2.32 (1H, m), 2.35–2.42 (3H, m), 2.80 (1H, m), 3.49 (1 H, dd,  $J=4.9$ , 9.5 Hz), 3.66 (1 H, dd,  $J=4.3$ , 9.5 Hz), 3.80 (3 H, s), 3.81 (3H, s), 4.05 (1H, m), 4.24 (1H, brs), 4.45 (2H, s), 4.84 (1H, d, J= 9.2 Hz), 5.68 (1H, dd, J=8.5, 15.0 Hz), 5.71 (1H, d, J=15.0 Hz), 6.07 (1H, dd, J=11.0, 15.0 Hz), 6.16 (1H, dd, J=11.3, 14.6 Hz), 6.21 (1H, d,  $J=7.3$  Hz), 6.43–6.49 (3H, m), 7.18 (1H, d,  $J=8.6$  Hz), 7.19 ppm (1H, dd, J = 11.3, 15.0 Hz); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  = 12.1, 16.2, 19.5, 21.1, 21.4, 22.3, 28.3, 29.7, 30.5, 31.7, 34.1, 34.9, 42.5, 44.0, 46.2, 48.3, 53.0, 55.4, 60.7, 68.1, 71.0, 76.8, 98.6, 104.0, 118.5, 122.4, 127.8, 128.2, 130.9, 132.1, 133.1, 140.4, 141.4, 145.5, 158.7, 160.9, 166.1, 209.8 ppm; FAB-MS  $m/z$ : 638  $[M^+ + H]$ ; HR-FAB-MS: calcd for  $C_{38}H_{56}NO_7$   $[M^+ + H]$ : 638.4057; found 638.4065.

# Total Synthesis of Scyphostatin **FULL PAPER**

(4S,5S,6S)-4,5-Epoxy-6-hydroxy-6-{(2S)-3-[(2,4-dimethoxyphenyl)methyloxy]-2-[(2E,4E,6E,12E)-(8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}-2-cyclohexen-1-one (35 b): LDA in THF  $(0.25 \text{ m})$  was prepared as follows: *nBuLi*  $(1.56 \text{ m} \text{ in hexane}, 0.78 \text{ mL})$  was added to a solution of diisopropylamine (70.3 µL, 0.5 mmol) in dry THF  $(1.15 \text{ mL})$  at  $-78 \text{°C}$  for 30 min. under Ar. LDA  $(0.25 \text{ m})$  in THF, 0.35 mL) and [15]crown-5 (69  $\mu$ L, 0.348 mmol) were added to a solution of 34b  $(7.4 \text{ mg}, 0.0116 \text{ mmol})$  in THF  $(0.3 \text{ mL})$  at  $-78 \text{ °C}$  under Ar. After being stirred for 10 min, a solution of N-tert-butylbenzenesulfinimidoyl chloride (75.1 mg, 0.348 mmol) in THF (0.3 mL) was added to the reaction mixture. After being stirred for 15 min, the reaction mixture was quenched with water and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using  $Et<sub>2</sub>O$  as the eluent to give 35b  $(2.6 \text{ mg}, 0.0058 \text{ mmol}, 35\%)$  as a yellow oil, and the substrate 34b  $(4.3 \text{ mg}, 58\%)$  was recovered.  $[\alpha]_D^{23} = +24.8$   $(c=0.30, \text{CHCl}_3)$ ; IR (KBr):  $\tilde{v} = 3271, 2958, 1697, 1651, 1612, 1514, 1461 \text{ cm}^{-1};$  <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta = 0.80$  (3H, d,  $J = 6.1$  Hz), 0.81 (3H, m), 0.90 (3H, d,  $J =$ 6.7 Hz), 0.99 (3H, d, J=6.7 Hz), 1.02 (1H, m), 1.16–1.33 (3H, m), 1.61  $(3H, s)$ , 1.74 (1H, m), 1.88 (1H, dd,  $J=6.7$ , 12.8 Hz), 2.01 (2H, m), 2.20 (1H, m), 2.32 (1H, m), 3.50 (1H, dd, J=4.9, 9.5 Hz), 3.52 (1H, m), 3.61  $(1\,\text{H}, \text{dd}, J = 3.7, 9.5 \text{ Hz})$ , 3.69  $(1\,\text{H}, \text{d}, J = 3.7 \text{ Hz})$ , 3.80  $(3\,\text{H}, \text{s})$ , 3.81  $(3\,\text{H}, \text{s})$ s), 4.22 (1H, m), 4.45 (2H, s), 4.84 (1H, d, J=9.2 Hz), 5.68 (1H, dd, J= 9.8, 14.7 Hz), 5.69 (1H, d, J=14.7 Hz), 5.97 (1H, d, J=9.0 Hz), 6.04–6.21  $(3H, m), 6.41-6.49$   $(3H, m), 7.07$   $(1H, dd, J=4.3, 9.0 Hz), 7.17$   $(1H, d,$  $J=9.2$  Hz), 7.21 ppm (1H, dd,  $J=11.6$ , 14.7 Hz); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  = 12.1, 19.5, 21.1, 21.4, 28.3, 29.7, 30.5, 31.6, 34.1, 35.0, 38.3, 44.0, 45.5, 47.9, 48.3, 53.0, 55.5, 56.3, 68.3, 71.8, 98.6, 114.0, 118.5, 122.2, 127.8, 128.3, 130.5, 130.8, 132.1, 133.1, 140.5, 144.1, 144.6, 145.6, 158.9, 160.9, 165.9, 197.7 ppm; FAB-MS m/z: 636 [M++H]; HR-FAB-MS: calcd for  $C_{38}H_{54}NO_7$  [ $M^+$ +H]: 636.3900; found 636.3876.

(4S,5S,6S)-4,5-Epoxy-6-hydroxy-6-{(2S)-3-hydroxy-2-[(2E,4E,6E,12E)- (8R,10S,14R)-8,10,12,14-(tetramethyl)hexadeca-2,4,6,12-tetraenoylamino]propyl}-2-cyclohexen-1-one (scyphostatin (1)):  $Ph_3C^{+}BF_4^-$  (1.8 mg,

 $0.00564$  mmol) was added to a solution of  $35b$  (3.0 mg, 0.0047 mmol) in  $CH_2Cl_2$  (0.2 mL) at 0 °C under Ar. After being stirred for 5 min, the reaction mixture was quenched with saturated  $NAHCO<sub>3</sub>$  (aq.) and extracted with EtOAc. The organic layer was washed with brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and evaporated in vacuo. The residue was purified by column chromatography using EtOAc/acetone (4:1) as the eluent to give scyphostatin (1; 1.5 mg, 0.00312 mmol, 66%).  $[\alpha]_D^{25} = +60.5$  (c=0.20, MeOH); IR (KBr):  $\tilde{v} = 3292, 2956, 2923, 1697, 1651, 1606, 1512,$ 1454 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD):  $\delta$  = 0.83–0.86 (6H, m), 0.91  $(3H, d, J=6.7 \text{ Hz})$ , 1.00  $(3H, d, J=6.8 \text{ Hz})$ , 1.04  $(1H, m)$ , 1.19  $(1H, m)$ , 1.27–1.41 (2H, m), 1.54 (3H, s), 1.58 (1H, m), 1.78 (1H, m), 1.86 (1H, m), 1.88 (1H, m), 2.08 (1H, dd, J=3.7, 14.7 Hz), 2.27 (1H, m), 2.35 (1H, m), 3.46 (1H, dd, J=5.5, 11.0 Hz), 3.52 (1H, dd, J=4.9, 11.0 Hz), 3.59  $(1\,\text{H}, \text{m})$ , 3.66  $(1\,\text{H}, \text{d}, J=3.7\,\text{Hz})$ , 4.04  $(1\,\text{H}, \text{m})$ , 4.86  $(1\,\text{H}, \text{m})$ , 5.71  $(1\,\text{H},$ m), 5.89(1H, d, J=15.3 Hz), 6.07 (1H, dd, J=1.3, 9.8 Hz), 6.15 (1H, dd,  $J=11.6$ , 14.6 Hz), 6.26 (1H, dd,  $J=11.0$ , 15.3 Hz), 6.54 (1H, dd,  $J=11.0$ , 14.6 Hz), 7.09–7.18 ppm (2H, m); <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>OD):  $\delta$  = 12.9, 16.9, 20.5, 22.0, 22.5, 30.1, 32.2, 36.0, 36.7, 40.3, 46.0, 48.5, 49.8, 50.1, 58.8, 65.9, 78.0, 124.2, 129.9, 130.4, 132.6, 134.5, 134.7, 141.8, 142.9, 146.4, 146.8, 169.0, 200.4 ppm; FAB-MS m/z: 486 [M++H]; HR-FAB-MS m/z: 486.3199 (calcd for  $C_{29}H_{43}NO_5H$ : 486.3219); the <sup>1</sup>H and <sup>13</sup>C NMR, IR, and HR-FAB-MS spectra showed good agreement with those of the authentic sample.

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Sports, Science, and Technology, Japan and the Shorai Foundation for Science and Technology.

- [1] For the structure determination of scyphostatin including the absolute configurations of the cyclohexanone unit and biological activity, see: a) M. Tanaka, F. Nara, K. Suzuki, T. Hosoya, T. Ogita, [J. Am.](http://dx.doi.org/10.1021/ja9713385) [Chem. Soc.](http://dx.doi.org/10.1021/ja9713385) 1997, 119[, 7871 – 7872](http://dx.doi.org/10.1021/ja9713385); b) F. Nara, M. Tanaka, T. Hosoya, K. Suzuki, T. Ogita, J. Antibiot. 1999, 52, 525; c) F. Nara, M. Tanaka, S. Masuda-Inoue, Y. Yamamoto, H. Doi-Yoshioka, K. Suzuki-Konagai, S. Kumakura, T. Ogita, J. Antibiot. 1999, 52, 531; for the determination of the absolute configurations of the side chain, see: d) S. Saito, N. Tanaka, K. Fujimoto, H. Kogen, [Org. Lett.](http://dx.doi.org/10.1021/ol991351p)  $2000, 2, 505 - 506.$
- [2] For examples of recently reported N-SMase inhibitors, see: a) C. Arenz, A. Giannis, [Angew. Chem.](http://dx.doi.org/10.1002/(SICI)1521-3757(20000417)112:8%3C1498::AID-ANGE1498%3E3.0.CO;2-Y) 2000, 112, 1498-1500; [Angew.](http://dx.doi.org/10.1002/(SICI)1521-3773(20000417)39:8%3C1440::AID-ANIE1440%3E3.0.CO;2-R) [Chem. Int. Ed.](http://dx.doi.org/10.1002/(SICI)1521-3773(20000417)39:8%3C1440::AID-ANIE1440%3E3.0.CO;2-R) 2000, 39, 1440 – 1442; b) C. Arenz, A. Giannis, [Eur. J.](http://dx.doi.org/10.1002/1099-0690(200101)2001:1%3C137::AID-EJOC137%3E3.0.CO;2-%23) [Org. Chem.](http://dx.doi.org/10.1002/1099-0690(200101)2001:1%3C137::AID-EJOC137%3E3.0.CO;2-%23) 2001, 137 – 140; c) C. Arenz, M. Thutewohl, O. Block, H. Waldmann, H.-J. Altenbach, A. Giannis, [ChemBioChem](http://dx.doi.org/10.1002/1439-7633(20010202)2:2%3C141::AID-CBIC141%3E3.0.CO;2-P) 2001, 2, [141 – 143](http://dx.doi.org/10.1002/1439-7633(20010202)2:2%3C141::AID-CBIC141%3E3.0.CO;2-P); d) C. Arenz, M. Gartner, V. Wascholowski, A. Giannis, [Bioorg. Med. Chem.](http://dx.doi.org/10.1016/S0968-0896(01)00165-1) 2001, 9, 2901 – 2904; e) T. Yokomatsu, H. Takechi, T. Akiyama, S. Shibuya, T. Kominato, S. Soeda, H. Shimeno, [Bioorg. Med. Chem. Lett.](http://dx.doi.org/10.1016/S0960-894X(01)00179-2) 2001, 11, 1277-1280; f) T. Hakogi, Y. Monden, M. Taichi, S. Iwama, S. Fujii, K. Ikeda, S. Katsumura, [J.](http://dx.doi.org/10.1021/jo025529o) [Org. Chem.](http://dx.doi.org/10.1021/jo025529o) 2002, 67[, 4839– 4846](http://dx.doi.org/10.1021/jo025529o); g) C. C. Lindsey, C. Gomez-Diza, J. M. Villalba, T. R. R. Pettus, [Tetrahedron](http://dx.doi.org/10.1016/S0040-4020(02)00391-5) 2002, 58[, 4559– 4565](http://dx.doi.org/10.1016/S0040-4020(02)00391-5); h) T. Yokomatsu, T. Murano, T. Akiyama, J. Koizumi, S. Shibuya, Y. Tsuji, S. Soeda, H. Shimeno, [Bioorg. Med. Chem. Lett.](http://dx.doi.org/10.1016/S0960-894X(02)00888-0) 2003, 13, [229– 236](http://dx.doi.org/10.1016/S0960-894X(02)00888-0); i) M. Taguchi, K. Sugimoto, K. Goda, T. Akama, K. Yamamoto, T. Suzuki, Y. Tomishima, M. Nishiguchi, K. Arai, K. Takahashi, T. Kobori, Bioorg. Med. Chem. Lett. 2003, 13, 1963 – 1966; j) M. Taguchi, K. Goda, K. Sugimoto, T. Akama, K. Yamamoto, T. Suzuki, Y. Tomishima, M. Nishiguchi, K. Arai, K. Takahashi, T. Kobori, [Bioorg. Med. Chem. Lett.](http://dx.doi.org/10.1016/j.bmcl.2003.08.020) 2003, 13, 3681 – 3684; k) S. Soeda, Y. Tsuji, T. Ochiai, K. Mishima, K. Iwasaki, M. Fujiwara, T. Yokomatsu, T. Murano, S. Shibuya, H. Shimeno, Tetrahedron 2004, 45, 619– 626.
- [3] S. Chatterjee, Arterioscler. Thromb. Vasc. Biol. 1998, 18, 1523-1533.
- [4] For a synthetic study of the cyclohexenone segment in its optically active form, see: a) T. Izuhara, T. Katoh, [Tetrahedron Lett.](http://dx.doi.org/10.1016/S0040-4039(00)01312-5) 2000, 41, [7651 – 7655](http://dx.doi.org/10.1016/S0040-4039(00)01312-5); b) T. Izuhara, T. Katoh, [Org. Lett.](http://dx.doi.org/10.1021/ol015873s) 2001, 3[, 1653 – 1656](http://dx.doi.org/10.1021/ol015873s); c) M. K. Gurjar, S. Hotha, Heterocycles 2000, 53, 1885 – 1889; d) L. M. Murray, P. O'Brien, R. J. K. Taylor, [Org. Lett.](http://dx.doi.org/10.1021/ol034521d)  $2003$ , 5, [1943 – 1946](http://dx.doi.org/10.1021/ol034521d); e) T. Katoh, T. Izuhara, W. Yokota, M. Inoue, K. Watanabe, A. Nobeyama, T. Suzuki, [Tetrahedron](http://dx.doi.org/10.1016/j.tet.2005.10.082) 2006, 62[, 1590 – 1608](http://dx.doi.org/10.1016/j.tet.2005.10.082); for a synthetic study of the cyclohexenone segment in its nonoptically active form, see: f) K. A. Runcie, R. J. K. Taylor, [Org. Lett.](http://dx.doi.org/10.1021/ol0164132) 2001, 3[, 3237 – 3239](http://dx.doi.org/10.1021/ol0164132); g) R. Takagi, W. Miyanaga, Y. Tamura, K. Ohkata, [Chem. Commun.](http://dx.doi.org/10.1039/b206163e) 2002, 2096 – 2097; h) R. Takagi, K. Tojo, M. Iwata, K. Ohkata, [Org. Biomol. Chem.](http://dx.doi.org/10.1039/b504039f) 2005, 3, 2031 – 2036; i) E. N. Pitsinos, A. Cruz, [Org. Lett.](http://dx.doi.org/10.1021/ol0506359) 2005, 7[, 2245 – 2248](http://dx.doi.org/10.1021/ol0506359); j) N. G. Stevenson, C. D. Savi, J. P. Harrity, Synlett 2006, 2272 – 2274.
- [5] For an alternative study on the synthesis and configuration of the scyphostatin side chain, see: a) T. R. Hoye, M. A. Tennakoon, Org. Lett. 2000, 2, 1481-1483; b) R. Takagi, S. Tsuyumine, H. Nishitani, W. Miyanaga, K. Ohkata, [Aust. J. Chem.](http://dx.doi.org/10.1071/CH03298) 2004, 57, 439– 447; c) Z. Tan, E. Negishi, [Angew. Chem.](http://dx.doi.org/10.1002/ange.200353429) 2004, 116, 2971 – 2974; [Angew.](http://dx.doi.org/10.1002/anie.200353429) [Chem. Int. Ed.](http://dx.doi.org/10.1002/anie.200353429) 2004, 43, 2911 – 2914; d) G. D. McAllister, R. J. K. Taylor, [Tetrahedron Lett.](http://dx.doi.org/10.1016/j.tetlet.2004.01.156) 2004, 45, 2551 – 2554.
- [6] a) M. Inoue, W. Yokota, M. G. Murugesh, T. Izuhara, T. Katoh, [Angew. Chem.](http://dx.doi.org/10.1002/ange.200454192) 2004, 116, 4303 – 4305; [Angew. Chem. Int. Ed.](http://dx.doi.org/10.1002/anie.200454192) 2004, 43[, 4207 – 4209](http://dx.doi.org/10.1002/anie.200454192); b) M. Inoue, W. Yokota, T. Katoh, Synthesis 2007,  $622 - 637$ .
- [7] R. Takagi, W. Miyanaga, K. Tojo, S. Tsuyumine, K. Ohkata, [J. Org.](http://dx.doi.org/10.1021/jo070337k) [Chem.](http://dx.doi.org/10.1021/jo070337k) 2007, 72[, 4117 – 4125](http://dx.doi.org/10.1021/jo070337k).
- [8] H. Fujioka, N. Kotoku, Y. Sawama, Y. Nagatomi, Y. Kita, [Tetrahe](http://dx.doi.org/10.1016/S0040-4039(02)00916-4)[dron Lett.](http://dx.doi.org/10.1016/S0040-4039(02)00916-4) 2002, 43[, 4825 – 4828.](http://dx.doi.org/10.1016/S0040-4039(02)00916-4)
- [9] H. Fujioka, N. Kotoku, Y. Sawama, H. Kitagawa, Y. Ohba, T-L. Wang, Y. Nagatomi, Y. Kita, [Chem. Pharm. Bull.](http://dx.doi.org/10.1248/cpb.53.952) 2005, 53, 952 – 957.
- [10] According to the procedure of D. Scholz, U. Schmidt, [Chem. Ber.](http://dx.doi.org/10.1002/cber.19741070714) 1974, 107[, 2295 – 2298](http://dx.doi.org/10.1002/cber.19741070714), benzoic acid was converted into cyclohexa-2,5-dienylethanal diethylacetal in four steps.
- [11] The synthesis of stannyl compound 19 is given in the Supporting In-formation; W. C. Still, [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja00473a025) 1978, 100, 1481-1487.
- [12] a) I. Ohtani, T. Kusumi, Y. Kashman, H. Kakisawa, [J. Am. Chem.](http://dx.doi.org/10.1021/ja00011a006) Soc. 1991, 113[, 4092 – 4096](http://dx.doi.org/10.1021/ja00011a006); b) T. Kusumi, T. Yabuuchi, H. Takahashi, T. Ooi, J. Synth. Org. Chem. Jpn. 2005, 63, 1102 – 1114.
- [13] B. Lal, B. N. Pramanik, M. S. Manhas, A. K. Bose, [Tetrahedron Lett.](http://dx.doi.org/10.1016/S0040-4039(01)83657-1) 1977, 18[, 1977 – 1980.](http://dx.doi.org/10.1016/S0040-4039(01)83657-1)
- [14] K. B. Sharpless, R. C. Michaelson, [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja00799a061) 1973, 95, [6136 – 6137](http://dx.doi.org/10.1021/ja00799a061).
- [15] a) H. Fujioka, Y. Sawama, N. Murata, T. Okitsu, O. Kubo, S. Matsuda, Y. Kita, [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja046103p) 2004, 126, 11800 – 11801; b) H. Fujioka, T. Okitsu, Y. Sawama, N. Murata, R. Li, Y. Kita, [J. Am. Chem.](http://dx.doi.org/10.1021/ja060328d) Soc. 2006, 128, 5930-5938; TMSOTf is also effective for the deprotection of acetal moieties with acidic labile functional groups as well as TESOTf.
- [16] W. P. Griffith, S. V. Ley, G. P. Whitcombe, A. P. White, [J. Chem. Soc.](http://dx.doi.org/10.1039/c39870001625) [Chem. Commun.](http://dx.doi.org/10.1039/c39870001625) 1987, 1625 – 1627.
- [17] a) T. Mukaiyama, J. Matsuo, [Chem. Lett.](http://dx.doi.org/10.1246/cl.2000.1250) 2000[, 1250 1251](http://dx.doi.org/10.1246/cl.2000.1250); b) T. Mukaiyama, J. Matsuo, M. Yanagisawa, [Chem. Lett.](http://dx.doi.org/10.1246/cl.2000.1072) 2000[, 1072 – 1075.](http://dx.doi.org/10.1246/cl.2000.1072)
- [18] D. H. R. Barton, P. D. Magnus, G. Streckert, D. Zurr, Chem. Commun. 1971, 1109–1110.
- [19] a) T. W. Greene, P. G. Wuts, Protective Groups in Organic Synthesis, 3rd ed., Wiley, New York, 1999; b) J. R. Hanson, Protecting Groups in Organic Synthesis, 1st ed.; Blackwell Science Inc., Malden, 1999.
- [20] T. Onoda, R. Sirai, S. Iwasaki, [Tetrahedron Lett.](http://dx.doi.org/10.1016/S0040-4039(97)00043-9) 1997, 38, 1443 [1446.](http://dx.doi.org/10.1016/S0040-4039(97)00043-9)
- [21] a) D. B. Dess, J. C. Martin, [J. Org. Chem.](http://dx.doi.org/10.1021/jo00170a070) 1983, 48, 4155-4156; b) D. B. Dess, J. C. Martin, [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja00019a027) 1991, 113, 7277 – 7287.

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